*	*	*	*	*	* *	* *	* Welcome to STN International * * * * * * * * * *
1	NEW	S	1				Web Page for STN Seminar Schedule - N. America
N	NEW	S	2		DEC	01	ChemPort single article sales feature unavailable
ľ	NEW	S	3		FEB	02	Simultaneous left and right truncation (SLART) added for CERAB, COMPUAB, ELCOM, and SOLIDSTATE
N	VEW	S	4		FEB	02	GENBANK enhanced with SET PLURALS and SET SPELLING
1	VEW	S	5		FEB	06	Patent sequence location (PSL) data added to USGENE
	NEW		6		FEB		COMPENDEX reloaded and enhanced
	VEW		7		FEB	11	WTEXTILES reloaded and enhanced
	NEW		8		FEB		New patent-examiner citations in 300,000 CA/CAplus
							patent records provide insights into related prior art
1	NEW	S	9		FEB	19	Increase the precision of your patent queries use terms from the IPC Thesaurus, Version 2009.01
ľ	VEW	S	10		FEB	23	Several formats for image display and print options discontinued in USPATFULL and USPAT2
N	NEW	S	11		FEB	2.3	MEDLINE now offers more precise author group fields
-		~					and 2009 MeSH terms
N	NEW	S	12		FEB	23	TOXCENTER updates mirror those of MEDLINE - more
_		_				_	precise author group fields and 2009 MeSH terms
1	NEW	S	13		FEB	23	Three million new patent records blast AEROSPACE into
							STN patent clusters
ľ	NEW	S	14		FEB	25	USGENE enhanced with patent family and legal status
							display data from INPADOCDB
ľ	NEW	S	15		MAR	06	INPADOCDB and INPAFAMDB enhanced with new display
							formats
N	VEW	S	16		MAR	11	EPFULL backfile enhanced with additional full-text
							applications and grants
ľ	NEW	S	17		MAR	11	ESBIOBASE reloaded and enhanced
ľ	VEW	S	18		MAR	20	CAS databases on STN enhanced with new super role
							for nanomaterial substances
ľ	VEW	S	19		MAR	23	CA/CAplus enhanced with more than 250,000 patent
							equivalents from China
ľ	VEW	S	20		MAR	30	IMSPATENTS reloaded and enhanced
ľ	VEW	S	21		APR	03	CAS coverage of exemplified prophetic substances
							enhanced
1	VEW	S	22		APR	07	STN is raising the limits on saved answers
ľ	VEW	S	23		APR	24	CA/CAplus now has more comprehensive patent assignee
							information
1	VEW	S	24		APR	26	USPATFULL and USPAT2 enhanced with patent
							assignment/reassignment information
	NEW				APR	28	CAS patent authority coverage expanded
	VEW				APR	28	ENCOMPLIT/ENCOMPLIT2 search fields enhanced
ľ	NEW	S	27		APR	28	Limits doubled for structure searching in CAS REGISTRY
ľ	NEW	S	28		MAY	8 0	STN Express, Version 8.4, now available
1	VEW	S	29		MAY	11	STN on the Web enhanced
ľ	VEW	S	30		MAY	11	BEILSTEIN substance information now available on
							STN Easy
ľ	VEW	S	31		MAY	14	DGENE, PCTGEN and USGENE enhanced with increased
							limits for exact sequence match searches and
							introduction of free HIT display format
ľ	VEW	S	32		MAY	15	INPADOCDB and INPAFAMDB enhanced with Chinese legal
							status data

NEWS EXPRESS JUNE 27 08 CURRENT WINDOWS VERSION IS V8.3, AND CURRENT DISCOVER FILE IS DATED 06 APRIL 2009.

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FILE LAST UPDATED: 21 May 2009 (20090521/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Feb 2009
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Feb 2009

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=> s 22811-02-5 or 10220-46-9 or 2917-26-2 or 2885-00-9 or thioglycolate or mercaptoacetate or hexadecanethiol or hexadecylthiol or mercaptan or octadecanethiol or octadecylthiol

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress...

Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L2 2011 L1

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Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L4 1621 L3

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Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

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2218 MERCAPTOACETATE

1405 HEXADECANETHIOL

41 HEXADECYLTHIOL

22248 MERCAPTAN

1668 OCTADECANETHIOL

91 OCTADECYLTHIOL

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DECANETHIOL OR HEXADECYLTHIOL OR MERCAPTAN OR OCTADECANETHIOL
OR OCTADECYLTHIOL

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The previous command name entered was not recognized by the system. For a list of commands available to you in the current file, enter "HELP COMMANDS" at an arrow prompt (=>).

=> s 111 and 112

L13 1 L11 AND L12

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L13 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

AN 2003:798402 CAPLUS

DN 139:311931

TI Metal coating of hair fibers for cosmetics

IN Vic, Gabin; Livoreil, Aude; Giroud, Franck

PA L'oreal, Fr.

SO Fr. Demande, 18 pp.

CODEN: FRXXBL

DT Patent

LA French

FAN.CNT 1

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PRAI	FR	FR 2002-4352				Α	2	0020	0408										
	US 2002-372455P					P	2	0020	0416										

RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

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L13 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

AN 2003:798402 CAPLUS

DN 139:311931

ED Entered STN: 12 Oct 2003

TI Metal coating of hair fibers for cosmetics

IN Vic, Gabin; Livoreil, Aude; Giroud, Franck

PA L'oreal, Fr.

SO Fr. Demande, 18 pp.

CODEN: FRXXBL

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AΒ
     The invention relates to a treatment process which confers cosmetic
     properties on hair fibers. The process consists of treating the fibers
     with a metal salt in the presence of a reducing agent, directly on the
     fiber to form the corresponding free metal. Thus, a lock of hair after
     being shampooed, was dried and an aq. soln. of AgNO3 was applied onto the
     hair. After the addn. of NaBH4, the natural pigmented hair was dark, with
     metallic brilliance reflected on it.
ST
    metal salt hair cosmetic
    Alcohols, biological studies
ΙT
```

RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);

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PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (C1-4; metal treatment of hair fibers for
        cosmetics)
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     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
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        cosmetics)
ΤТ
     Polyelectrolytes
       Surfactants
        (amphoteric; metal treatment of hair fibers for
        cosmetics)
ΤT
     Fats and Glyceridic oils, biological studies
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        cosmetics)
ΙT
     Surfactants
        (anionic; metal treatment of hair fibers
        for cosmetics)
    Polyelectrolytes
ΤT
       Surfactants
        (cationic; metal treatment of hair fibers for
        cosmetics)
ΤТ
    Cosmetics
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ΙT
     Sulfates, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (hydrogen; metal treatment of hair fibers for
        cosmetics)
    Antifoaming agents
ΤТ
     Antiperspirants
     Cosmetics
     Hair
     Hair preparations
     Perfumes
     Pigments, nonbiological
     Preservatives
     Reducing agents
     Shampoos
     Sunscreens
     Thickening agents
        (metal treatment of hair fibers for cosmetics)
ΙT
     Alkaline earth salts
     Bromates
     Carbonates, biological studies
     Disulfides
     Halides
     Nitrates, biological studies
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     Polymers, biological studies
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     Proteins
     Rare earth salts
     Sulfates, biological studies
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Thioethers
     Thiosulfates
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     Vitamins
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
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ΙT
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     Enzymes, reactions
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        (metal treatment of hair fibers for cosmetics)
ΙT
        (moisturizers; metal treatment of hair fibers for
        cosmetics)
ΙT
     Surfactants
        (nonionic; metal treatment of hair fibers
        for cosmetics)
ΙT
     Peroxysulfates
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (peroxymonosulfates; metal treatment of hair fibers
        for cosmetics)
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    Alcohols, biological studies
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ΤT
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ΤТ
     Thiols, reactions
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ΙT
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     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (vegetable; metal treatment of hair fibers for
        cosmetics)
     64-17-5, Ethanol, biological studies 67-63-0, Isopropanol, biological
ΙT
             67-64-1, Acetone, biological studies 78-93-3, Methyl ethyl
```

ketone, biological studies 79-20-9, Methyl acetate 110-71-4 123-86-4, Butyl acetate 141-78-6, EtOAc, biological studies 7429-90-5D, Aluminum, salts 7439-89-6D, Iron, salts 7439-98-7D, Molybdenum, salts 7440-02-0D, Nickel, salts 7440-05-3D, Palladium, salts 7440-06-4D, Platinum, salts 7440-22-4D, Silver, salts 7440-31-5D, Tin, salts 7440-32-6D, Titanium, salts 7440-33-7D, Tungsten, salts 7440-36-0D, Antimony, salts 7440-50-8D, Copper, salts 7440-57-5D, Gold, salts 7440-66-6D, Zinc, salts 7440-74-6D, Indium, salts 7758-89-6, Cuprous chloride 7761-88-8, Silver nitrate, biological studies 7775-41-9, Silver fluoride 7783-89-3, Silver bromate 7783-90-6, Silver chloride, biological studies 7783-96-2, Silver iodide 7785-23-1, Silver bromide 7787-70-4, Cuprous bromide 10025-98-6, Dipotassium palladium tetrachloride 10294-26-5, Silver 10294-28-7, Gold tribromide 16903-35-8 16923-58-3, Disodium hexachloroplatinate 19045-66-0D, Thiocarbamic acid, salts 73506-93-1, Diethoxyethane RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (Uses) (metal treatment of hair fibers for cosmetics) 50-81-7, Ascorbic acid, reactions 53-57-6, NaDPH 58-68-4, NaDH 68-11-1, Thioglycolic acid, reactions 77-92-9D, Citric acid, salts 106-51-4, 2,5-Cyclohexadiene-1,4-dione, reactions 123-31-9, Hydroquinone, reactions 280-64-8, 9-BBN 1758-73-2, Formamidinesulfinic acid 2885-00-9, 1-Octadecanethiol 3483-12-3, Dithiothreitol 6838-83-1, Diisoamylborane 7772-98-7 7775-14-6 7803-51-2, Phosphine 13762-51-1 14451-43-5 16853-85-3 16940-66-2 17836-88-3 25895-60-7, Sodium cyanoborohydride 37318-49-3, Protein disulfide isomerase 56553-60-7 131760-67-3 145626-87-5 RL: RCT (Reactant); RACT (Reactant or reagent) (metal treatment of hair fibers for cosmetics) RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD (1) Anon; PATENT ABSTRACTS OF JAPAN 1991, V015(243), PC-0842 (2) Anon; PATENT ABSTRACTS OF JAPAN 1992, V016(509), PC-0997 (3) Dong Sung Pharmaceuticals Co L; EP 1243249 A 2002 CAPLUS (4) Grollier, J; US 4971596 A 1990 CAPLUS (5) Kanebo Ltd; JP 03077806 A 1991 CAPLUS (6) Katsumi, M; JP 04187625 A 1992 CAPLUS (7) Lapidus, H; US 4195972 A 1980 (8) Richez, H; US 1055355 A 1913 CAPLUS (9) Wella Aq; DE 2806603 A 1979 CAPLUS (10) Wella Aq; DE 29621557 U 1997 => s silver or ag 379544 SILVER 342362 AG 522045 SILVER OR AG => d his (FILE 'HOME' ENTERED AT 00:05:08 ON 23 MAY 2009) FILE 'CAPLUS' ENTERED AT 00:05:27 ON 23 MAY 2009

S 22811-02-5/REG# OR 10220-46-9/REG# OR 2917-26-2/REG# OR 28

ΙT

L14

L1

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L8
             6 S L7
L9
         32684 S L8 OR L6 OR L4 OR L2 OR THIOGLYCOLATE OR MERCAPTOACETATE OR H
         84695 S (NONIONIC OR NON-IONIC OR ANIONIC OR ZWITTERIONIC) AND SURFAC
L10
L11
           319 S L9 AND L10
          6680 S METAL TREAT##### OR TREAT###### METAL
L12
L13
             1 S L11 AND L12
L14 522045 S SILVER OR AG
=> 111 \text{ and } 114
L11 IS NOT A RECOGNIZED COMMAND
The previous command name entered was not recognized by the system.
For a list of commands available to you in the current file, enter
"HELP COMMANDS" at an arrow prompt (=>).
=> s 111 and 114
     12 L11 AND L14
L15
=> d 1-12 all
L15 ANSWER 1 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
AN
    2008:1046144 CAPLUS
DN
    149:312963
ED Entered STN: 29 Aug 2008
TI Preparation of conductive supported noble metal nanoparticle catalysts
IN Stucky, Galen D.; Zheng, Nanfeng
    The Regents of the University of California, USA
PΑ
SO
    U.S. Pat. Appl. Publ., 35pp.
    CODEN: USXXCO
DT
    Patent
    English
INCL 428403000; 216055000; 428402000; 264005000; 264007000; 5021000000;
     502300000; 502159000; 502355000; 502350000
     56-4 (Nonferrous Metals and Alloys)
     Section cross-reference(s): 57, 67
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                         APPLICATION NO. DATE
                       ____
                                          _____
     _____
                                                                 _____
PI US 20080206562 A1 20080828 PRAI US 2007-884668P P 20070112
                                         US 2008-13436
                                                                20080112
CLASS
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PATENT NO.
               CLASS PATENT FAMILY CLASSIFICATION CODES
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US 20080206562 INCL
                       428403000; 216055000; 428402000; 264005000; 264007000;
                       502100000; 502300000; 502159000; 502355000; 502350000
                       B32B0015-02 [I,A]; C23F0001-00 [I,A]; B29B0009-00
                IPCI
                       [I,A]; B29B0009-16 [I,A]; B01J0031-06 [I,A];
                       B01J0021-04 [I,A]; B01J0021-08 [I,A]; B01J0023-34
                       [I,A]; B01J0023-16 [I,C*]; B01J0029-00 [I,A];
                       B01J0021-18 [I,A]; B01J0021-00 [I,C*]; B01J0027-06
                       [I,A]; B01J0023-42 [I,A]; B01J0023-44 [I,A];
                       B01J0023-50 [I,A]; B01J0023-52 [I,A]; B01J0023-48
                       [I,C*]; B01J0027-02 [I,A]; B01J0027-24 [I,A];
                       B01J0031-02 [I,A]; B01J0023-755 [I,A]; B01J0031-26
                       [I,A]
                       428/403.000; 216/055.000; 216/083.000; 264/005.000;
                NCL
                       264/007.000; 428/402.000; 502/080.000; 502/087.000;
                       502/100.000; 502/150.000; 502/159.000; 502/167.000;
                       502/168.000; 502/171.000; 502/180.000; 502/181.000;
                       502/200.000; 502/216.000; 502/232.000; 502/300.000;
                       502/325.000; 502/337.000; 502/339.000; 502/340.000;
                       502/344.000; 502/345.000; 502/347.000; 502/349.000;
                       502/350.000; 502/355.000
    The prepn. of elec.-conductive noble metal nanoparticle catalysts on
AΒ
    catalyst supports such as alumina, silica, titania, clays, zeolites, or
    carbon black, is described.
ST
    gold silver palladium nanocatalyst support sol gel micelle ceramic
ΙT
    Solvents
        (aprotic; prepn. of conductive supported noble metal nanoparticle
       catalysts)
    Polyethers, uses
ΙT
    RL: MOA (Modifier or additive use); USES (Uses)
        (arom., alkyl-, surfactants; prepn. of conductive supported
       noble metal nanoparticle catalysts)
ΙT
    Thiols, uses
    RL: MOA (Modifier or additive use); USES (Uses)
       (caps on catalyst nanoparticles; prepn. of conductive supported noble
       metal nanoparticle catalysts)
ΤТ
    Bentonite, processes
    Carbon black, processes
    Clays, processes
    Diatomite
    Silica gel, processes
    Zeolites (synthetic), processes
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
    engineered material use); PROC (Process); USES (Uses)
        (catalyst supports; prepn. of conductive supported noble metal
       nanoparticle catalysts)
TΤ
    Nanoparticles
        (catalysts; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΙT
    Alcohols, uses
    RL: MOA (Modifier or additive use); USES (Uses)
        (ethoxylated, surfactants; prepn. of conductive supported
       noble metal nanoparticle catalysts)
    Hydrocarbons, processes
ΤТ
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
    engineered material use); PROC (Process); USES (Uses)
        (fluoro, catalyst supports; prepn. of conductive supported noble metal
       nanoparticle catalysts)
    Surfactants
ΤТ
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(in coatings; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΙT
     Electroluminescent devices
    Molecular electronic devices
     Optoelectronics
     Secondary batteries
     Semiconductor devices
     Sensors
     Solar cells
        (nanocatalysts for; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     Photolysis catalysts
        (nanocatalysts; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΤT
     Catalysts
     Semiconductor materials
        (nanoparticles; prepn. of conductive supported noble metal nanoparticle
        catalysts)
     Surfactants
ΤT
        (nonionic; prepn. of conductive supported noble metal
        nanoparticle catalysts)
     Silsesquioxanes
ΙT
     RL: RGT (Reagent); RACT (Reactant or reagent)
        (octyl- and hexyl-; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     Dyes
        (org.-, functional mol.; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     Calcination
     Catalyst supports
     Etching
     Reducing agents
        (prepn. of conductive supported noble metal nanoparticle catalysts)
ΙT
     7440-44-0, Carbon, processes
     RL: PEP (Physical, engineering or chemical process); TEM (Technical or
     engineered material use); PROC (Process); USES (Uses)
        (activated, catalyst supports; prepn. of conductive supported noble
        metal nanoparticle catalysts)
     64-17-5, Ethanol, uses
ΙT
     RL: NUU (Other use, unclassified); USES (Uses)
        (buffer soln.; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΙT
     49543-63-7, 4-(tert-Butyl)benzyl mercaptan
     RL: MOA (Modifier or additive use); USES (Uses)
        (cap on nanoparticles; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     1322-36-7, Dodecanethiol
     RL: MOA (Modifier or additive use); USES (Uses)
        (caps on catalyst nanoparticles; prepn. of conductive supported noble
        metal nanoparticle catalysts)
ΙT
     7440-02-0P, Nickel, preparation
                                     7440-05-3P, Palladium, preparation
     7440-06-4P, Platinum, preparation 7440-22-4P, Silver,
                   7440-50-8P, Copper, preparation 7440-57-5P, Gold,
     preparation
     preparation 12006-51-8P, AuCu
     RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation);
     USES (Uses)
        (catalyst nanoparticles; prepn. of conductive supported noble metal
        nanoparticle catalysts)
     1309-48-4, Magnesium oxide (MgO), processes 1314-23-4, Zirconia,
ΙT
               1344-28-1, Aluminum oxide (Al2O3), processes 7631-86-9,
```

ΙT

ΙT

ΙT

ΙT

ΙT

IT

TΤ

ΙT

ΤТ

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Silica, processes 7782-42-5, Graphite, processes 13463-67-7, Titania,
processes
RL: PEP (Physical, engineering or chemical process); TEM (Technical or
engineered material use); PROC (Process); USES (Uses)
   (catalyst supports; prepn. of conductive supported noble metal
  nanoparticle catalysts)
1306-38-3, Cerium oxide (CeO2), uses
                                     1313-13-9, Manganese oxide (MnO2),
uses 1313-96-8, Niobium oxide (Nb2O5)
RL: MOA (Modifier or additive use); USES (Uses)
   (coatings on colloidal silica; prepn. of conductive supported noble
  metal nanoparticle catalysts)
12638-19-6P
RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation);
USES (Uses)
   (nanoparticles; prepn. of conductive supported noble metal nanoparticle
  catalysts)
2966-50-9, Silver trifluoroacetate 14024-17-0, Iron acetyl
          14024-61-4 14024-64-7 16902-59-3 17927-72-9
                                                             19443-16-4
acetonate
                       23894-03-3 24772-51-8 27858-32-8, Titanium
19443-17-5
           23894-00-0
diisopropoxide bis(ethyl acetoacetate) 62905-51-5
                                                    65574-21-2
65583-10-0 66197-44-2 82269-80-5 93918-06-0, Aluminum sec-butoxide
bis(ethyl acetoacetate) 98719-26-7 140190-96-1 144665-26-9
204522-78-1 299957-41-8 380240-62-0 1050499-47-2 1050499-48-3
1050499 - 49 - 4 1050499 - 50 - 7 1050499 - 51 - 8 1050499 - 52 - 9 1050499 - 53 - 0
1050499-54-1
RL: RCT (Reactant); RACT (Reactant or reagent)
   (precursors; prepn. of conductive supported noble metal nanoparticle
  catalysts)
1313-99-1, Nickel oxide, uses
                               1345-25-1, Ferrous oxide, uses
11104-61-3, Cobalt oxide
RL: MOA (Modifier or additive use); USES (Uses)
   (prepn. of conductive supported noble metal nanoparticle catalysts)
78-07-9, Ethyltriethoxysilane 78-10-4, Tetraethoxysilane
Tetramethoxysilane 682-01-9, Tetrapropoxysilane
                                                  1185-55-3,
Methyltrimethoxysilane
                       1336-21-6, Ammonium hydroxide ((NH4)(OH))
2031-67-6, Methyltriethoxysilane 4766-57-8, Tetrabutoxysilane
30232-12-3 192082-40-9, Mercaptoundecanoic acid
RL: RGT (Reagent); RACT (Reactant or reagent)
   (prepn. of conductive supported noble metal nanoparticle catalysts)
1722-26-5, Triethylamine-borane
                               4856-95-5
                                             7337-45-3,
tert-Butylamine-borane 13774-81-7, Ammonia-borane
RL: RGT (Reagent); RACT (Reactant or reagent)
   (reducing agents; prepn. of conductive supported noble metal
  nanoparticle catalysts)
                           71-43-2, Benzene, uses 75-09-2,
67-66-3, Chloroform, uses
Dichloromethane, uses 108-88-3, Toluene, uses 110-54-3, Hexane, uses
110-82-7, Cyclohexane, uses
RL: NUU (Other use, unclassified); USES (Uses)
   (solvent; prepn. of conductive supported noble metal nanoparticle
  catalysts)
14243-64-2
RL: PEP (Physical, engineering or chemical process); PROC (Process)
   (substrates; prepn. of conductive supported noble metal nanoparticle
   catalysts)
577-11-7, Sodium bis(2-ethylhexyl) sulfosuccinate 9002-89-5, Polyvinyl
alcohol 9002-92-0, Brij 30 9004-98-2, Brij 97 9036-19-5,
(Octylphenoxy)polyethoxyethanol 12441-09-7D, Sorbitan, ester derivs.
27251-32-7
RL: MOA (Modifier or additive use); USES (Uses)
   (surfactants; prepn. of conductive supported noble metal
```

nanoparticle catalysts)

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L15 ANSWER 2 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
    2007:415891 CAPLUS
ΑN
DN
   146:463862
ED
  Entered STN: 16 Apr 2007
TI Discoloration prevention of metals using organic ultra-thin films and
    methods therefor
IN Liang, Chenghao; Yang, Changjiang; Huang, Naibao
PA Dalian Maritime University, Peop. Rep. China
  Faming Zhuanli Shenging Gongkai Shuomingshu, 10pp.
SO
    CODEN: CNXXEV
DТ
    Patent.
LA Chinese
CC
    42-10 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 46, 56
FAN.CNT 1
                                      APPLICATION NO.
    PATENT NO.
                     KIND DATE
                                                            DATE
                                       _____
                      ____
  CN 1943882
                     A 20070411 CN 2006-10134093 20061026
PΤ
PRAI CN 2006-10134093
                            20061026
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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[I,A]; C23C0022-05 [I,A]; C07C0321-04 [I,A];
                     C07C0321-00 [I,C*]
               IPCR B05D0007-14 [I,C]; B05D0007-14 [I,A]
OS
    MARPAT 146:463862
AB
    Film-forming solns. contain 0.001-1 mol/L alkyl thiols and 0.001-1 mol/L
    surfactants. Thus, a coating soln. on Ag contained stearyl thiol 15,
    polyethylene glycol nonylphenyl ether 7, hexadecyltrimethylammonium
    bromide 2, Pluronic 64 7 g/L.
    metal discoloration prevention coating surfactant thiol; silver
ST
    discoloration prevention coating surfactant thiol
    Surfactants
ΤТ
       (anionic; coating materials contg. thiols and
       surfactants for discoloration prevention of metals)
ΙT
    Surfactants
       (cationic; coating materials contg. thiols and surfactants
       for discoloration prevention of metals)
ΙT
    Discoloration prevention
       (coating materials contg. thiols and surfactants for
       discoloration prevention of metals)
ΤT
    Quaternary ammonium compounds, uses
    Thiols, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
       (coating materials contg. thiols and surfactants for
       discoloration prevention of metals)
ΙT
    Coating materials
       (discoloration-resistant; coating materials contg. thiols and
       surfactants for discoloration prevention of metals)
    691397-13-4, Pluronic L 64
ΤТ
    RL: TEM (Technical or engineered material use); USES (Uses)
       (Pluronic L 64; coating materials contg. thiols and surfactants
       for discoloration prevention of metals)
    57-09-0, Hexadecyltrimethylammonium bromide 2885-00-9,
    Stearylmercaptan 7440-22-4, Silver, uses 9016-45-9,
```

Polyethylene glycol nonylphenyl ether

RL: TEM (Technical or engineered material use); USES (Uses) (coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

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L15 ANSWER 3 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
ΑN
   2007:150387 CAPLUS
DN
   146:236227
ED
   Entered STN: 09 Feb 2007
ΤI
    Conductive adhesive composition comprising pressure sensitive adhesive and
    electrolyte
    Menon, Vinod P.; Kumar, Kanta; Nelson, Carl T.; Rizzardi, Don A.
ΙN
    3M Innovative Properties Company, USA
PΑ
    U.S. Pat. Appl. Publ., 20pp.
SO
    CODEN: USXXCO
DT
    Patent
   English
T.A
INCL 600391000; 600392000; 252500000
   63-7 (Pharmaceuticals)
FAN.CNT 1
                   KIND DATE
                                        APPLICATION NO.
                                                                DATE
    PATENT NO.
                       ____
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    US 20070032719 A1 20070208 US 2005-197216
AU 2006278717 A1 20070215 AU 2006-278717
    AU 2006278717
                                                                 20060801
    CA 2617273 A1 20070215 CA 2006-2617273 20060801
WO 2007019115 A1 20070215 WO 2006-US29794 20060801
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
            CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
            GE, GH, GM, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP,
            KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN,
            MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU,
            SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG,
            US, UZ, VC, VN, ZA, ZM, ZW
         RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
            IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,
            CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,
            GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
            KG, KZ, MD, RU, TJ, TM
                        A1 20080507 EP 2006-789019
     EP 1917318
                                                                  20060801
         R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR
     JP 2009503235 T 20090129 JP 2008-525088 20060801
    MX 2008001425
                        A
                              20080416 MX 2008-1425
                                                                 20080129
                             20080508 KR 2008-702725
     KR 2008040689
                       A
                       A 20080806 CN 2006-80028822
A 20081128 IN 2008-CN571
A 20050804
W 20060801
     CN 101238189
                                                                 20080204
IN 2008CN00571
PRAI US 2005-197216
WO 2006-US29794
                                                                 20080204
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 US 20070032719 INCL 600391000; 600392000; 252500000
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                       A61B0005-04 [I,A]; H01B0001-12 [I,A]; H01B0001-00 [I,A]
                IPCR
                       A61B0005-04 [I,C]; A61B0005-04 [I,A]; H01B0001-00
                       [I,C]; H01B0001-00 [I,A]; H01B0001-12 [I,C];
                       H01B0001-12 [I,A]
                NCL
                       600/391.000; 252/500.000; 600/392.000
                       C09J009/02; A61B005/0408F; A61N001/04; C09J133/04+B4;
                ECLA
                       C09J133/06+B2; H01B001/20; K61B; M08L; M08L; M08L
                IPCI
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
 AU 2006278717
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IPCR
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                        C09J009/02; A61B005/0408F; A61N001/04; C09J133/04+B4;
                        C09J133/06+B2; H01B001/20; K61B; M08L; M08L; M08L
CA 2617273
                 IPCI
                        A61B0005-0408 [I,A]; A61B0018-14 [I,A]; A61K0050-00
                        [I,A]; A61N0001-04 [I,A]; C09J0009-02 [I,A];
                        C09J0009-00 [I,C*]; C09J0011-06 [I,A]; C09J0011-02
                        [I,C*]
                 IPCR
                        C09J0009-00 [I,C]; C09J0009-02 [I,A]; A61B0005-0408
                        [I,C]; A61B0005-0408 [I,A]; A61B0018-14 [I,C];
                        A61B0018-14 [I,A]; A61K0050-00 [I,C]; A61K0050-00
                        [I,A]; A61N0001-04 [I,C]; A61N0001-04 [I,A];
                        C09J0011-02 [I,C]; C09J0011-06 [I,A]
 WO 2007019115
                 IPCI
                        C09J0009-02 [I,A]; C09J0009-00 [I,C*]
                 TPCR
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
                 ECLA
                       C09J009/02; A61B005/0408F; A61N001/04; C09J133/04+B4;
                        C09J133/06+B2; H01B001/20; K61B; M08L; M08L; M08L
                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
 EP 1917318
                 IPCI
                 IPCR
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
JP 2009503235
                 IPCI
                       C09J0201-00 [I,A]; C09J0009-02 [I,A]; C09J0009-00
                        [I,C*]; C09J0004-02 [I,A]; A61L0024-00 [I,A];
                        A61N0001-04 [I,A]
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                        4C081/CA071; 4C081/CA081; 4C081/CA101; 4C081/CA16;
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                        4J040/JA03; 4J040/JB09; 4J040/KA12; 4J040/KA13;
                        4J040/KA32; 4J040/KA38; 4J040/KA39; 4J040/MA14;
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                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
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 IN 2008CN00571 IPCI
                        C09J0009-02 [ICM, 7]; C09J0009-00 [ICM, 7, C*]
OS
    MARPAT 146:236227
    A conductive adhesive compn. is provided and articles that include the
AΒ
     adhesive compn. as a component thereof. The conductive adhesive compn.
     comprises: (a) pressure sensitive adhesive; (b) electrolyte comprising
     water sol. or water dispersible orq. chloride; and (c) humectant. In some
     embodiments, the conductive adhesive compn. is a bicontinuous compn.
     comprising an aq. phase and an oil phase, and the bicontinuous compn. may
     be derived from a polymerizable microemulsion compn., the microemulsion
     compn. comprising: an aq. phase comprising one or more hydrophilic
     monomers or oligomers and/or one or more amphiphilic monomers or oligomers
     in water, the water-sol. or water-dispersible org. chloride, surfactant
     and humectant; and an oil phase comprising one or more hydrophobic
     monomers or oligomers. Biomedical articles such as biomedical electrodes,
     may incorporate the foregoing adhesive as a component. For example,
     adhesive precursor comprised of acrylic acid 15 g, 2-hydroxyethyl
     methacrylate 20 g, tetrakis(hydroxymethyl)phosphonium chloride 11 g,
     1,3-butylene glycol 25 g, glycerol 10 g, water 19 g, Irgacure 2959 0.55 g
     and polyethylene glycol diacrylate 0.15 g. The precursor was coated using
     a knife coater onto a release liner as substrate. The knife was set so
     that a 25 mil (0.64 mm) thick coating was obtained. Polymn. was induced
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in the coated microemulsion by exposure to UV radiation. A total dose of

 $1800~\mathrm{mJ/cm2}$ was applied over approx. $7~\mathrm{min}$, forming a conductive, bicontinuous adhesive. This conductive adhesive had an excellent adhesion to human skin.

- ST polymer acrylate electrolyte chloride conductive adhesive
- IT Polyurethanes, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (acrylates; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Electric conductors

(adhesive; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Fats and Glyceridic oils, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (almond, amidopropalkonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Surfactants

(anionic; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Fats and Glyceridic oils, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (avocado, amidopropalkonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Surfactants

(cationic; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Onium compounds

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Quaternary ammonium compounds, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (chlorides; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Fatty acids, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (coco, trimethylammonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Chain transfer agents

Crosslinking agents

Electrodes

Electrolytes

Human

Humectants

Hvdroaels

Surfactants

(conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Alcohols, uses

Thiols, uses

RL: NUU (Other use, unclassified); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Acrylic polymers, biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

IT Sulfonium compounds

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

- IT Adhesives
 - (conductive; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT Soybean oil
 - RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (dimethylammonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT Surfactants
 - (nonionic; conductive adhesive compn. comprising pressure
 sensitive adhesive and electrolyte)
- IT Chlorides, biological studies
 - RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (org.; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT Adhesives
 - (pressure-sensitive; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT Fatty acids, biological studies
 - RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (tallow, bishydroxyethyl/dime quaternary ammonium compds.; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT 558-13-4, Carbon tetrabromide 25103-09-7, Isooctyl **thioglycolate** , uses
 - RL: NUU (Other use, unclassified); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- IT 1070-70-8, 1,4-Butanediol diacrylate 1321-74-0, Divinylbenzene, reactions 10526-04-2, 1,8-Octanediol diacrylate 13048-33-4, 1,6-Hexanediol diacrylate
 - RL: RCT (Reactant); RACT (Reactant or reagent) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- 56-34-8, Tetraethylammonium chloride 56-37-1, Benzyltriethylammonium chloride 56-81-5, Glycerin, biological studies 56-93-9, Benzyltrimethylammonium chloride 57-55-6, Propylene glycol, biological 77-99-6, Trimethylolpropane 88-12-0D, polymer studies 67-48-1 107-21-1, Ethylene glycol, biological studies 107-88-0, 1,3-Butanediol 110-63-4, 1,4-Butanediol, biological studies 112-00-5, Dodecyltrimethylammonium chloride 112-02-7, Hexadecyltrimethylammonium chloride 112-03-8, Octadecyltrimethylammonium chloride 124-64-1, Tetrakis(hydroxymethyl)phosphonium chloride 139-08-2, Tetradecyldimethylbenzylammonium chloride 593-81-7D, Trimethylammonium chloride, coco fatty acid derivs. 7173-51-5 9004-98-2, Brij 98 17301-53-0, Behenyltrimethylammonium chloride Phosphonium chloride 25265-71-8, Dipropylene glycol 26570-48-9, Polyethylene oxide diacrylate 26597-36-4 32862-91-2, Oxonium chloride 60182-11-8, Polyethylene glycol acrylate 93507-51-8 106797-53-9,

IRGACURE 2959 123776-56-7 145687-02-1, Pemulen TR 2

923929-97-9 923929-99-1 924299-17-2, Hetoxol OL 35
RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)

463965-14-2

- IT 7783-90-6, **Silver** chloride, biological studies
 - RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (conductive ink soln.; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte)
- L15 ANSWER 4 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
- Full Text
- AN 2005:1062684 CAPLUS

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143:351549
DN
     Entered STN: 05 Oct 2005
ED
     Water-based sulfur-containing composition chemical mechanical polishing of
TΙ
     nonferrous metals
     Johns, Peter Gamon; Harrison, Clare Elizabeth
IN
     Middlesex Silver Co. Limited, UK
PΑ
     Brit. UK Pat. Appl., 29 pp.
SO
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     English
     ICM C23F011-16
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     ICS C23F011-00
CC
     57-7 (Ceramics)
     Section cross-reference(s): 56
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                  A 20070411 CN 2005-80013434
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                        T 20071220 JP 2007-505641 20050324
A 20070713 IN 2006-DN5356 20060915
A 20061116 MX 2006-10964 20060925
A1 20071206 US 2007-594477 20070702
     JP 2007537354
IN 2006DN05356

MX 2006010964

US 20070277906

PRAI GB 2004-7163

WO 2005 0751
                         A 20040330
     WO 2005-GB50043 W
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CLASS
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                        C23F011-16
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 MX 2006010964 IPCI
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                        148/022.000
OS
    MARPAT 143:351549
    A compn. and assocd. method of manuf. of a water based compn. comprising a
AΒ
    treatment agent selected from an alkanethiol, alkyl thioglycollate, and
     dialkyl sulfide or dialkyl disulfide. The compn. also includes at least
     one of an amphoteric, non-ionic or cationic surfactant, where the
     treatment agent is directly dissolved or dispersed the water contg. the
     amphoteric, non-ionic or cationic surfactant. The compn. is
     particularly useful for the treatment of Ag-Cu-Ge alloy, copper, brass,
     and nickel. A solid polishing medium can also be included in the compn.,
     for example, silica or pptd. chalk, alumina, or silica.
     chalk alumina silica alkanethiol thioglycollate chem mech polishing copper
ST
     Thiols, processes
ΤТ
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
        (alkanethiol; water-based sulfur-contq. compn. chem. mech. polishing of
        metals)
     Disulfides
ΤТ
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
        (alkyl; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΤТ
     Chalk
     Diatomite
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΙT
     Surfactants
        (cationic; water-based sulfur-contg. compn. chem. mech. polishing of
ΙT
     Polishina
        (chem.-mech.; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
     Polishing materials
ΙT
        (paste; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΙT
     Thioethers
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (water-based sulfur-contg. compn. chem. mech. polishing of metals)
     1344-28-1, Alumina, uses
ΤТ
     RL: TEM (Technical or engineered material use); USES (Uses)
        (abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
     9004-82-4, Sodium laureth sulfate
ΙT
     RL: MOA (Modifier or additive use); USES (Uses)
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```
(anionic surfactant; water-based sulfur-contg.
        compn. chem. mech. polishing of metals)
ΙT
     7631-86-9, Silica, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΙT
     36574-66-0D, N-coco acyl derivs.
     RL: MOA (Modifier or additive use); USES (Uses)
        (cocamidopropyl betaine, surfactant; water-based
        sulfur-contg. compn. chem. mech. polishing of metals)
     7440-02-0, Nickel, processes
                                   7440-50-8, Copper, processes
ΙT
                                                                   11144-43-7
     12597-71-6, Brass, processes
                                    74969-69-0
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); PROC (Process)
        (polished substrate; water-based sulfur-contg. compn. chem. mech.
       polishing of metals)
ΙT
     62-56-6, Thiourea, uses 2885-00-9, Octadecyl mercaptan
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing compn. component; water-based sulfur-contg. compn. chem.
       mech. polishing of metals)
IT 2917-26-2, Hexadecyl mercaptan
     RL: MOA (Modifier or additive use); USES (Uses)
        (surfactant; water-based sulfur-contg. compn. chem. mech.
       polishing of metals)
     68-11-1D, alkyl esters
ΤT
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (water-based sulfur-contg. compn. chem. mech. polishing of metals)
RE.CNT 6
             THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Anon; EP 0492487 A1 CAPLUS
(2) Anon; GB 0956927 A
(3) Anon; GB 1117510 A
(4) Anon; US 3503883 A
(5) Anon; US 3518098 A
(6) Anon; US 5650385 A CAPLUS
L15 ANSWER 5 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
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    143:295501
ED
    Entered STN: 19 Jul 2005
TΙ
     Single Etch Patterning of Stacked Silver and Molybdenum Alloy Layers on
    Glass Using Microcontact Wave Printing
ΑU
    Burdinski, Dirk; Brans, Harold J. A.; Decre, Michel M. J.
CS
    Philips Research, Eindhoven, 5656 AA, Neth.
SO
    Journal of the American Chemical Society (2005), 127(31), 10786-10787
    CODEN: JACSAT; ISSN: 0002-7863
PΒ
    American Chemical Society
DT
    Journal
LA
    English
    74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Section cross-reference(s): 56
    Stacked thin layers of silver alloy (AgPdCu) and MoCr layers on 10
AB
     	imes 15 cm2 glass substrates were patterned by microcontact wave
     printing and etching. Patterns of etch-resistant octadecanethiol
     self-assembled monolayers (SAMs) were wave printed with regular backplane
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stabilized PDMS stamps. Pattern development was achieved by etching both metal layers in a single step, employing a nitric acid-based etching bath. Trifluoroacetic acid and a nitrite salt were identified as essential bath components for a homogeneous etching process. Etch defects could be eliminated by the addn. of a decanesulfonate, which stabilizes the SAM resist via a defect healing mechanism.

- ST etching **silver** molybdenum alloy electrode display
- IT Liquid crystal displays

(active matrix; single etch patterning of stacked **silver** and molybdenum alloy layers on glass using microcontact wave printing as electrodes for)

IT Surfactants

ΤT

(anionic; single etch patterning of stacked silver
 and molybdenum alloy layers on glass using microcontact wave printing)
Lithography

(microcontact printing; single etch patterning of stacked
silver and molybdenum alloy layers on glass using microcontact
wave printing)

IT Autocatalysis

Electrodes

Etching

Glass substrates

Self-assembled monolayers

(single etch patterning of stacked ${\tt silver}$ and molybdenum alloy layers on glass using microcontact wave printing)

IT 64-19-7, Acetic acid, processes 76-05-1, Trifluoroacetic acid, processes 7632-00-0, Sodium nitrite 7664-38-2, Phosphoric acid, processes 7697-37-2, Nitric acid, processes

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(etchant; single etch patterning of stacked **silver** and molybdenum alloy layers on glass using microcontact wave printing)

IT 2885-00-9, 1-Octadecanethiol

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(ink, self-assembled monolayer; single etch patterning of stacked **silver** and molybdenum alloy layers on glass using microcontact wave printing)

IT 188820-19-1 317855-00-8

RL: CPS (Chemical process); DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (single etch patterning of stacked **silver** and molybdenum

alloy layers on glass using microcontact wave printing)

IT 13419-61-9, Sodium decane sulfonate

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(**surfactant** for etching soln.; single etch patterning of stacked **silver** and molybdenum alloy layers on glass using microcontact wave printing)

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Full Text
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AN
    141:353637
DN
ED
    Entered STN: 15 Oct 2004
    Pretreatment of Ag-alloy surface with organosulfur compounds for
    tarnishing prevention
    Johns, Peter Gammon; Harrison, Clare Elizabeth
ΙN
PΑ
    Middlesex Silver Co. Limited, UK
SO
    PCT Int. Appl., 43 pp.
    CODEN: PIXXD2
DT
    Patent
    English
LA
IC
    ICM C23F011-16
CC
    56-6 (Nonferrous Metals and Alloys)
FAN.CNT 1
                       NIND DATE
                                          ADDITOARTONI NO
P
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ΡI	WO	2004		A1 20		2004	0041014		WO 2004-GB1373					20040330				
		W:	ΑE,	AG,	AL,	AM,	AT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,
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			GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KΡ,	KR,	KΖ,	LC,
			LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NΙ,
			NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,
			ТJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW
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	AU	2004	2256	93		A1		2004	1014		AU 2	004-	2256	93		2	0040	330
	CA	2520	807			A1		2004	1014		CA 2	004-	2520	807		2	0040	330
	EP	1611	267			A1		2006	0104		EP 2	004-	7243	13		2	0040	330
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	CN	1780	937			Α		2006	0531		CN 2	004-	8001	1375		2	0040	330

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JP 2006523266 T 20061012 JP 2006-506057
IN 2005DN04346 A 20070831 IN 2005-DN4346
MX 2005010452 A 20060510 MX 2005-10452
US 20070039665 A1 20070222 US 2005-551476
PRAI GB 2003-7290 A 20030331
WO 2004-GB1373 W 20040330
                                                                 20040330
                                                                20050926
                                                                20050928
                                                                 20050929
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 WO 2004087996 ICM C23F011-16
                IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
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 AU 2004225693
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               ECLA C23F011/16; C23F011/16B
 CA 2520807
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               ECLA C23F011/16; C23F011/16B
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 EP 1611267
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               ECLA C23F011/16; C23F011/16B
 CN 1780937
               IPCI C23F0011-16 [I,A]; C23F0011-10 [I,C*]
               ECLA C23F011/16; C23F011/16B
 IPCR C23F0011-00 [I,C]; C23F0011-00 [I,A]; C22C0005-06
                       [I,C]; C22C0005-06 [I,A]; C22C0005-08 [I,A];
                       C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                FTERM 4K062/AA01; 4K062/BB21; 4K062/BC22; 4K062/FA16
 IN 2005DN04346 IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
 ECLA C23F011/16; C23F011/16B
 US 20070039665 IPCI C23G0001-00 [I,A]; C23C0022-58 [I,A]; C23C0022-05
                       [I,C*]
                       148/271.000; 134/002.000
                NCL
     The Ag alloys contg. minor Ge (esp. Ag-Cu-Ge alloys) to decrease the
AB
     fire stain discoloration are pretreated on the surface with an
     alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide
     to prevent tarnishing. The treatment with organosulfur compds. is
     suitable for manufd. Ag-alloy articles to prevent tarnished appearance
     during transit and the subsequent extended display without special
     packaging. The Ag-alloy surface is optionally treated with ag. soln.
     contg. an alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl
     disulfide, as well as a mixt. of anionic surfactant and amphoteric or
     nonionic surfactant to solubilize the treatment agent. The typical
     ternary alloy contains \mathbf{Ag} 80-96, Cu 1-19.9, and Ge 0.1-5%.
ST
     silver copper germanium alloy tarnishing prevention organosulfur
ΙT
     Surfactants
        (anionic, in tarnishing prevention; Ag-alloy
        surface treated with organosulfur compds. for tarnishing prevention)
ΙT
        (in tarnishing prevention; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
ΙT
     Surfactants
        (nonionic, in tarnishing prevention; Ag-alloy
        surface treated with organosulfur compds. for tarnishing prevention)
ΙT
        (prevention of; Aq-alloy surface treated with organosulfur
        compds. for tarnishing prevention)
```

ΙT

Thioethers

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Thiols, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (tarnishing prevention by; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
     7440-56-4, Germanium, uses
ΤТ
     RL: MOA (Modifier or additive use); USES (Uses)
        (Ag alloys contg., tarnishing prevention on; Ag
        -alloy surface treated with organosulfur compds. for tarnishing
       prevention)
     106-94-5, n-Propyl bromide
ΙT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (solvent, in tarnishing prevention; Ag-alloy surface treated
        with organosulfur compds. for tarnishing prevention)
IT 2885-00-9, Octadecyl mercaptan 2917-26-2,
     Cetyl mercaptan
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnishing prevention by; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
     39282-03-6, Sterling silver
                                   103221-24-5
                                                476614-10-5
     476614-12-7
                 476614-13-8
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnishing prevention on; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
ΙT
     9080-17-5, Ammonium polysulfide
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (test soln. with, for tarnishing; Ag-alloy surface treated
        with organosulfur compds. for tarnishing prevention)
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RE
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Full Text
AN
     2003:851241 CAPLUS
    139:330251
DN
ED
    Entered STN: 30 Oct 2003
ΤI
     Silver (carboxylate-n-alkyl thiolate) particles for photothermographic
     of thermographic imaging
     Ghyzel, Peter J.; Lelental, Mark; Dickinson, David A.; Pitt, Alan R.;
    Wear, Trevor J.
    Eastman Kodak Company, USA
PΑ
SO
    U.S., 14 pp.
    CODEN: USXXAM
DT
    Patent
LA
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IC
     ICM G03C001-498
INCL 430619000; 430611000; 430620000; 430631000
     74-2 (Radiation Chemistry, Photochemistry, and Photographic and Other
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FAN.CNT 1

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PATENT NO.
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                      B1 20031028 US 2002-200417
A1 20040128 EP 2003-77179
    US 6638708
                                                             20020722
PΤ
    EP 1385047
                                                              20030710
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    JP 2004054276 A 20040219 JP 2003-199297 20030718
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US 6638708
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                     [I,C*]; G03C0001-498 [I,A]
               NCL
                      430/619.000; 430/611.000; 430/620.000; 430/631.000
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EP 1385047
               IPCI G03C0001-498 [ICM, 7]
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               ECLA G03C001/498B; G03C001/498E1
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JP 2004054276
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                      2H123/AB25; 2H123/AB28; 2H123/BC00; 2H123/BC12;
                      2H123/CB00; 2H123/CB03
    The present disclosure relates to dispersions of {f silver}
AΒ
    (carboxylate-n-alkyl thiolate). The carboxylates are typically silver
    salts of long chain fatty acids and the n-alkyl thiolate is preferably
    1-dodecanethiol. These silver (carboxylate-n-alkyl thiolate) particles
    can be used to formulate imaging forming compns. that are useful in aq.
    thermog. or photothermog. imaging elements.
ST
    photog emulsion silver carboxylate alkyl thiolate particle photothermog
ΙT
    Photographic emulsions
       (heat-developable; silver (carboxylate-n-alkyl thiolate)
       particles for photothermog. of thermog. imaging)
ΤT
    Surfactants
       (nonionic; silver (carboxylate-n-alkyl thiolate)
       particles for photothermog. of thermog. imaging)
    Nanoparticles
ΤТ
       (silver (carboxylate-n-alkyl thiolate) particles for
       photothermog. of thermog. imaging)
ΤТ
    111-31-9, 1-Hexanethiol 112-55-0, 1-Dodecanethiol 112-85-6, Behenic
    acid 2885-00-9, 1-Octadecanethiol
    RL: TEM (Technical or engineered material use); USES (Uses)
       (silver (carboxylate-n-alkyl thiolate) particles for
       photothermog. of thermog. imaging)
RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
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   Silver n-Octadecanethiolate Layered Materials P2266
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(5) Voicu, R; Thermal Behavior of a Self-Assembled Silver n-Dode-canethiolate

L15 ANSWER 8 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

Layered Material Monitored by DSC P2642

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Full Text
AN 2003:798402 CAPLUS
DN
   139:311931
    Entered STN: 12 Oct 2003
ED
    Metal coating of hair fibers for cosmetics
ΤI
    Vic, Gabin; Livoreil, Aude; Giroud, Franck
ΙN
PΑ
   L'oreal, Fr.
   Fr. Demande, 18 pp.
SO
    CODEN: FRXXBL
DT
    Patent
LA
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    ICM A61K007-075
IC
CC
    62-3 (Essential Oils and Cosmetics)
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                      KIND DATE
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    FR 2838050
                       A1
                             20031010 FR 2002-4352
                                                                20020408
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    CN 1449737
                                         CN 2003-108449
                                                                 20030331
    CN 1213719 C 20050810
BR 2003000873 A 20040817 BR 2003-873
EP 1352630 A2 20031015 EP 2003-290860
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    EP 1352630
                       A3 20040324
    EP 1352630 B1 20060301
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    US 20030223944 A1 20031204 US 2003-407911 20030407 JP 2003300840 A 20031021 JP 2003-104420 20030408
JP 3759120
PRAI FR 2002-4352
                       B2 20060322
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[I,C*]; A61Q0001-02 [I,A]; A61Q0005-00 [I,C*];
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                       A61Q005/00; A61Q005/10
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                       A61K0007-06 [ICM, 7]
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                       A61Q005/12; A61K008/19; A61K008/27; A61K008/46;
                       A61Q005/00; A61Q005/10
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AB The invention relates to a treatment process which confers cosmetic

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Proteins

Rare earth salts

properties on hair fibers. The process consists of treating the fibers with a metal salt in the presence of a reducing agent, directly on the fiber to form the corresponding free metal. Thus, a lock of hair after being shampooed, was dried and an aq. soln. of AgNO3 was applied onto the hair. After the addn. of NaBH4, the natural pigmented hair was dark, with metallic brilliance reflected on it. metal salt hair cosmetic Alcohols, biological studies RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (Uses) (C1-4; metal treatment of hair fibers for cosmetics) Alkanes, biological studies RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (C5-10; metal treatment of hair fibers for cosmetics) Polyelectrolytes Surfactants (amphoteric; metal treatment of hair fibers for cosmetics) Fats and Glyceridic oils, biological studies RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (Uses) (animal; metal treatment of hair fibers for cosmetics) Surfactants (anionic; metal treatment of hair fibers for cosmetics) Polyelectrolytes Surfactants (cationic; metal treatment of hair fibers for cosmetics) Cosmetics (emollients; metal treatment of hair fibers for cosmetics) Sulfates, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (hydrogen; metal treatment of hair fibers for cosmetics) Antifoaming agents Antiperspirants Cosmetics Hair Hair preparations Perfumes Pigments, nonbiological Preservatives Reducing agents Shampoos Sunscreens Thickening agents (metal treatment of hair fibers for cosmetics) Alkaline earth salts Bromates Carbonates, biological studies Disulfides Halides Nitrates, biological studies Paraffin oils Phosphates, biological studies Polymers, biological studies Polysiloxanes, biological studies

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Sulfates, biological studies
     Thioethers
     Thiosulfates
     Transition metal salts
     Vitamins
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
        (metal treatment of hair fibers for cosmetics)
ΙT
     Bisulfites
     Enzymes, reactions
     Sulfites
     Thiols, reactions
     Thioredoxins
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (metal treatment of hair fibers for cosmetics)
ΙT
    Cosmetics
        (moisturizers; metal treatment of hair fibers for cosmetics)
ΤT
     Surfactants
        (nonionic; metal treatment of hair fibers for cosmetics)
    Peroxysulfates
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (peroxymonosulfates; metal treatment of hair fibers for cosmetics)
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ΤT
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
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     (Uses)
        (polyhydric; metal treatment of hair fibers for cosmetics)
     Sulfonic acids, biological studies
ΙT
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (salts; metal treatment of hair fibers for cosmetics)
ΙT
     Sulfinic acids
     Thiols, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (salts; metal treatment of hair fibers for cosmetics)
ΤТ
     Salts, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (thiol; metal treatment of hair fibers for cosmetics)
ΤТ
    Lactones
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
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     (Uses)
        (thiolactones; metal treatment of hair fibers for cosmetics)
     Fats and Glyceridic oils, biological studies
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     Tungsten, salts 7440-36-0D, Antimony, salts 7440-50-8D, Copper, salts
     7440-57-5D, Gold, salts 7440-66-6D, Zinc, salts 7440-74-6D, Indium,
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                       A61K0008-49 [I,A]; A61K0008-55 [I,A]; A61K0008-58
                       [I,A]; A61K0008-60 [I,A]; A61K0008-72 [I,C];
                       A61K0008-73 [I,A]; A61K0008-81 [I,A]; A61K0008-84
                       [I,A]; A61K0008-88 [I,A]; A61K0009-70 [I,C];
                       A61K0009-70 [I,A]; A61K0045-00 [I,C]; A61K0045-00
                       [I,A]; A61K0047-10 [I,C]; A61K0047-10 [I,A];
                       A61K0047-12 [I,C]; A61K0047-12 [I,A]; A61K0047-14
                       [I,C]; A61K0047-14 [I,A]; A61K0047-16 [I,C];
                       A61K0047-18 [I,A]; A61K0047-20 [I,C]; A61K0047-20
                       [I,A]; A61K0047-22 [I,C]; A61K0047-22 [I,A];
                       A61K0047-24 [I,C]; A61K0047-24 [I,A]; A61K0047-28
                       [I,C]; A61K0047-28 [I,A]; A61K0047-32 [I,C];
                       A61K0047-32 [I,A]; A61K0047-34 [I,C]; A61K0047-34
                       [I,A]; A61K0047-36 [I,C]; A61K0047-36 [I,A];
                       A61K0047-38 [I,C]; A61K0047-38 [I,A]; A61K0047-42
                       [I,C]; A61K0047-42 [I,A]; A61M0037-00 [I,C*];
                       A61M0037-00 [I,A]; A61Q0019-00 [I,C]; A61Q0019-00 [I,A]
                FTERM 4C076/AA72; 4C076/AA95; 4C076/BB31; 4C076/CC01;
                       4C076/CC03; 4C076/CC04; 4C076/CC18; 4C076/DD03;
                       4C076/DD04; 4C076/DD07; 4C076/DD08; 4C076/DD09;
                       4C076/DD13; 4C076/DD17; 4C076/DD38A; 4C076/DD66A;
                       4C076/EE06A; 4C076/EE10A; 4C076/EE12A; 4C076/EE13A;
                       4C076/EE17A; 4C076/EE23A; 4C076/EE26A; 4C076/EE27;
                       4C076/EE30A; 4C076/EE31A; 4C076/EE32A; 4C076/EE38A;
                       4C076/FF31; 4C076/FF35; 4C083/AA112; 4C083/AB032;
                       4C083/AC122; 4C083/AC131; 4C083/AC181; 4C083/AC371;
                       4C083/AC391; 4C083/AC421; 4C083/AC441; 4C083/AC532;
                       4C083/AC682; 4C083/AC772; 4C083/AC781; 4C083/AC791;
                       4C083/AD041; 4C083/AD042; 4C083/AD051; 4C083/AD071;
                       4C083/AD072; 4C083/AD091; 4C083/AD111; 4C083/AD131;
                       4C083/AD151; 4C083/AD201; 4C083/AD202; 4C083/AD211;
                       4C083/AD241; 4C083/AD261; 4C083/AD271; 4C083/AD281;
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4C083/AD282; 4C083/AD351; 4C083/AD391; 4C083/AD642; 4C083/AD662; 4C083/CC02; 4C083/DD12; 4C083/EE12; 4C083/EE13; 4C083/EE14; 4C083/EE16; 4C083/EE22; 4C084/AA17; 4C084/MA32; 4C084/MA63; 4C084/NA10; 4C084/ZA891
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- AΒ The present invention relates to a patch for controlled topical or transdermal delivery of effective levels of cosmetic, dermatol., and pharmaceutical active ingredients onto the skin, hair follicles, and sebaceous glands, with minimal discomfort and ease of use. The patch can be transparent or clear and comprises a rate-controlling matrix layer. The matrix layer comprises water-sensitive, bioadhesive, film forming polymers, a water sol. oligomer, and a surfactant. The cosmetic, dermatol., and pharmaceutical active ingredients are sol. or dispersed in the matrix. The patch becomes tacky when wetted and adheres onto the skin. The adhesive properties of the patch are sufficient to maintain the patch in place on the skin for the recommended treatment period while allowing the patch to be readily removed without causing skin irritation or leaving adhesive residue on the skin. For example, an antibiotic patch contained polyvinyl alc. 50, PVP 1, polysorbate 20 5, Maltrin 180 10, lactitol 5, glycerin 10, and chloramphenicol 0.55%.
- ST patch bioadhesive polymer oligosaccharide **surfactant**; antibiotic patch PVA PVP polysorbate chloramphenicol
- IT Glycosides

RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(alkyl polyglycosides; invisible patches contg. bioadhesive polymers and **surfactants**)

IT Surfactants

(amphoteric; invisible patches contg. bioadhesive polymers and surfactants)

IT Surfactants

(anionic; invisible patches contg. bioadhesive polymers and surfactants)

IT Surfactants

(cationic; invisible patches contg. bioadhesive polymers and surfactants)

IT Essential oils

RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(clove; invisible patches contg. bioadhesive polymers and surfactants)

IT Hair preparations

(conditioners; invisible patches contg. bioadhesive polymers and surfactants)

IT Cosmetics

(depilatories; invisible patches contg. bioadhesive polymers and surfactants)

IT Acne

Burn

Dandruff

Pruritus

Rhus diversiloba

Rhus toxicodendron

(drugs for; invisible patches contg. bioadhesive polymers and **surfactants**)

IT Alcohols, biological studies

Amides, biological studies

Esters, biological studies

RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

```
(ethoxylated; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
    Hair preparations
        (growth stimulants; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
    Vein, disease
        (hemorrhoid, drugs for; invisible patches contg. bioadhesive polymers
        and surfactants)
ΙT
     Syrups (sweetening agents)
        (hydrolyzed starch; invisible patches contq. bioadhesive polymers and
        surfactants)
     Allergy inhibitors
ΙT
     Aloe barbadensis
     Analgesics
     Anti-infective agents
     Anti-inflammatory agents
     Antibacterial agents
     Antibiotics
     Antiemetics
     Antihistamines
    Antimicrobial agents
    Antioxidants
    Antiperspirants
    Antitussives
     Antiviral agents
     Chelating agents
     Chemotherapy
     Cholinergic antagonists
     Deodorants
     Disinfectants
     Fungicides
     Hemostatics
     Immunomodulators
     Insecticides
     Radical scavengers
     Sunscreens
     Suntanning agents
     Vasoconstrictors
     Vasodilators
     Wound healing promoters
        (invisible patches contq. bioadhesive polymers and surfactants
        )
ΤT
    Amine oxides
     Amino acids, biological studies
     Carbohydrates, biological studies
     Caseins, biological studies
     Flavonoids
     Gelatins, biological studies
     Glycerides, biological studies
     Lanolin
     Lecithins
     Oligosaccharides, biological studies
     Paraffin oils
     Peptides, biological studies
     Polyamides, biological studies
     Polyesters, biological studies
     Polyoxyalkylenes, biological studies
     Polyoxyalkylenes, biological studies
     Polysaccharides, biological studies
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Proteins

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Retinoids
     Vitamins
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
ΙT
    Anesthetics
        (local; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Cosmetics
        (moisturizers; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (nonionic; invisible patches contg. bioadhesive polymers and
        surfactants)
     Amines, biological studies
ΤТ
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polyamines, nonpolymeric; invisible patches contg. bioadhesive
        polymers and surfactants)
    Alcohols, biological studies
ΙT
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polyhydric, propoxylated; invisible patches contq. bioadhesive
        polymers and surfactants)
     Quaternary ammonium compounds, biological studies
ΙT
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polymers; invisible patches contq. bioadhesive polymers and
        surfactants)
ΤТ
     Skin, disease
        (rash, drugs for; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Cosmetics
        (skin-lightening; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Drug delivery systems
        (tapes; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Cosmetics
        (wrinkle-preventing; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (zwitterionic; invisible patches contq. bioadhesive polymers
        and surfactants)
     36574-66-0D, N-coco acyl derivs.
ΤТ
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (cocoamidopropylbetaine; invisible patches contq. bioadhesive polymers
        and surfactants)
ΙT
     68-26-8, Retinol
                       96-26-4, Dihydroxyacetone
                                                   814-71-1, Calcium
     thioglycolate 34452-51-2, Potassium thioglycolate
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
     50-70-4, Sorbitol, biological studies
                                             50-70-4D, Sorbitol, oligomers
ΙT
            50-78-2, Aspirin 50-81-7, Vitamin C, biological studies
     50-99-7D, Glucose, esters 50-99-7D, D-Glucose, oligomers contg.
     55-56-1, Chlorhexidine 56-81-5, Glycerin, biological studies
                                                                     56-86-0D,
     Glutamic acid, N-acyl derivs. 57-48-7D, Fructose, oligomers contg.
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57-50-1D, Sucrose, esters 57-50-1D, Sucrose, oligomers contg. 57-55-6,
     Propylene glycol, biological studies 58-86-6D, Xylose, oligomers contg.
     59-23-4D, Galactose, oligomers contg. 59-87-0, Nitrofurazone 60-54-8,
     Tetracycline 69-65-8D, Mannitol, oligomers contg. 69-72-7, Salicylic
    acid, biological studies 69-79-4D, Maltose, oligomers contg. 87-99-0D, Xylitol, oligomers contg. 106-11-6, Diethylene glycol monostearate 107-36-8D, Isethionic acid, cocoyl derivs. 108-46-3, Resorcinol,
     biological studies 108-95-2, Phenol, biological studies 114-07-8,
     Erythromycin 115-83-3, Pentaerythritol tetrastearate 144-55-8, Sodium
     bicarbonate, biological studies 151-21-3, Sodium lauryl sulfate,
     biological studies 404-86-4, Capsaicin 497-19-8, Sodium carbonate,
     biological studies 585-86-4D, Lactitol, oligomers contg. 585-88-6D,
     Maltitol, oligomers contg. 770-35-4, Phenoxyisopropanol 1338-41-6,
     Sorbitan monostearate 1406-18-4, Vitamin E 2216-51-5 3380-34-5,
     Triclosan 3458-28-4D, D-Mannose, oligomers contg.
                                                           6284-40-8
     7439-97-6, Mercury, biological studies 7440-22-4, Silver,
     biological studies 7553-56-2, Iodine, biological studies 8011-96-9,
     Calamine 8050-81-5, Simethicone 9000-01-5, Gum arabic 9002-89-5,
    Polyvinyl alcohol 9002-98-6 9003-05-8, Polyacrylamide 9003-39-8,
     Polyvinylpyrrolidone 9004-64-2, Hydroxypropyl cellulose 9005-25-8,
     Starch, biological studies 9005-25-8D, Starch, hydrolyzates 9005-64-5,
     Polysorbate 20 9011-13-6, Styrene-maleic anhydride copolymer
     9011-16-9, Methyl vinyl ether-maleic anhydride copolymer 11099-07-3,
     Glycerin stearate 11111-12-9, Cephalosporin 11140-06-0, Glycerin
     palmitate 12694-22-3, Diglyceryl monostearate 13718-94-0D, Palatinose,
     oligomers contg. 15687-27-1, Ibuprofen 18323-44-9, Clindamycin
     25322-68-3, Polyethylene glycol 25322-69-4 25655-41-8, Povidone iodine
     26658-19-5, Sorbitan tristearate 27195-16-0, Sucrose distearate
     30233-64-8, Glyceryl monobehenate 39529-26-5, Decaglyceryl decastearate
     42852-72-2 53998-08-6, Sarcosinate 63119-59-5, Diglycerin distearate
     68424-04-4, Polydextrose 71185-87-0, Hexaglyceryl tristearate
     75537-01-8, Gantrez S-97 95461-64-6, Decaglyceryl pentastearate
     99734-29-9, Tetraglyceryl tristearate 99880-64-5, Glyceryl dibehenate
     106392-12-5, Polyoxyethylene polyoxypropylene block copolymer
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
     56-75-7, Chloramphenicol 94-09-7, Benzocaine
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (invisible patches contq. bioadhesive polymers and surfactants
L15 ANSWER 10 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
    1992:536001 CAPLUS
    117:136001
OREF 117:23503a,23506a
ED
    Entered STN: 04 Oct 1992
    Aqueous emulsion for temporary protection of silver and copper surfaces
     against tarnishing
IN
    Grossmann, Hermann
    Doduco GmbH und Co. Dr. Eugen Duerrwaechter, Germany
    Eur. Pat. Appl., 6 pp.
    CODEN: EPXXDW
    Patent
    German
     ICM C23F011-16
     56-10 (Nonferrous Metals and Alloys)
FAN.CNT 1
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ΙT

ΑN DN

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PATENT NO.
                      KIND DATE
                                        APPLICATION NO.
                                                             DATE
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                      A1 19920701 EP 1991-121903
PΙ
    EP 492487
                                                              19911220
                       B1 19960320
    EP 492487
       R: DE, ES, FR, GB, IT, NL
    DE 4041596 A1 19920702 DE 1990-4041596
ES 2086471 T3 19960701 ES 1991-121903
                                                              19901222
                                                               19911220
ES 2086471 T3 19960701
PRAI DE 1990-4041596 A 19901222
DE 1991-4124955 A 19910727
CLASS
           CLASS PATENT FAMILY CLASSIFICATION CODES
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EP 492487
                      C23F011-16
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               IPCI C23F0011-16 [ICM, 5]; C23F0011-10 [ICM, 5, C*]
               IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
               ECLA C23F011/16B
 DE 4041596
              IPCI C23F0011-12 [ICM,5]; C23F0011-16 [ICS,5]; C23F0011-10
                      [ICS,5,C*]; C09K0015-06 [ICA,5]; C09K0015-12 [ICA,5];
                      C09K0015-00 [ICA, 5, C*]; B01F0017-42 [ICA, 5];
                      B01F0017-38 [ICA,5]
                ECLA C23F011/16B
 ES 2086471
                IPCI C23F0011-16 [ICM, 6]; C23F0011-10 [ICM, 6, C*]
                IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                ECLA C23F011/16B
AΒ
    The emulsion of pH 1-10 (preferably 2-4) comprises a hydrophobic inhibitor
    of a C\geq 12 thioalc. with \geq 1 SH group and its ester 0.05-50
    (preferably 2-20), emulsifier 0.05-50 (2-20), and an anionic or
    nonionic surfactant \leq 2 (0.05-1 q/L). The emulsifier comprises
    an alkoxylated and preferably ethoxylated branched C4-20 alc., an alkyl or
    alkylphenyl ether of polyethylene glycol. Ag, Cu, and their alloys are
    treated with the emulsion at >T (m.p. of inhibitor), rinsed with H2O at
    <T, and dried with hot air. An example emulsion of pH 3 and suitable for
    treatment of Ag and Ag alloys contains octadecanethiol 0.5-30,
    polyethylene glycol alkyl ether 0.5-30, and SDS \leq 1 g/L H2O.
    tarnishing inhibitor silver copper; thiol SDS tarnishing inhibitor
ST
    silver; SDS thiol tarnishing inhibitor copper; polyethylene glycol ether
    tarnishing inhibitor
    Thiols, uses
ΙT
    RL: USES (Uses)
       (corrosion inhibitors, for copper and silver, with
       emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol)
ΤТ
    Tarnishing
       (of silver and copper alloys, aq. emulsion for prevention of)
    Corrosion inhibitors
ΙT
       (thiols, with emulsifiers of alkyl or alkyl Ph ether of polyethylene
       glycol)
ΙT
    Alcohols, compounds
    RL: PROC (Process)
       (C8-16, ethoxylated, corrosion inhibitor emulsion contg., thiol, for
       copper and silver and their alloys)
ΙT
    copper alloy, base
      silver alloy, base
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (tarnishing of, thiol inhibitor for)
    25322-68-3D, Polyethylene glycol, alkyl and alkylphenyl ethers 151-21-3,
ΙT
    RL: PROC (Process)
```

```
(corrosion inhibitor emulsion contg., thiol, for copper and
        silver and their alloys)
IT 2885-00-9, Octadecanethiol
     RL: PROC (Process)
        (corrosion inhibitors, for copper and silver, with
        emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol)
ΙT
     7440-22-4, Silver, reactions
                                   7440-50-8, Copper, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (tarnishing of, thiol inhibitor for)
L15 ANSWER 11 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
     1991:89162 CAPLUS
    114:89162
DN
OREF 114:15093a,15096a
    Entered STN: 09 Mar 1991
TΙ
     Silver metal liquidlike films (MELLFs). The effect of surfactants
ΑU
    Yogev, D.; Efrima, S.
     Dep. Chem., Ben-Gurion Univ. Negev, Beer-Sheva, 84105, Israel
CS
SO
    Langmuir (1991), 7(2), 267-71
    CODEN: LANGD5; ISSN: 0743-7463
DT
    Journal
    English
LA
CC
     66-4 (Surface Chemistry and Colloids)
     Section cross-reference(s): 73, 74
     The effects of surfactants on the prodn. and stabilization of Aq metal
AΒ
     liquidlike films (MELLFs) were studied. The main role of the surfactant
     is in stabilizing the Ag MELLFs and improving their properties
     (reflectivity, "fluidity"). A variety of different surfactants were
     found to be active, and of those investigated, anionic fluoroalkyl
     surfactants seem to be the most effective. In the case of anionic
     surfactants, the countercation has a significant effect on the Ag
     MELLF, esp. if it is a surface-active agent in itself. The effects of the
     surfactants on the interfacial tension and their effect on the measured
     reflectivities of the MELLFs are discussed in the context of the
     interfacial colloidal model of Ag MELLFs.
     silver metal liquidlike film formation; surfactant effect metal
    liquidlike film; interfacial tension metal liquidlike film
    Films
ΙT
        (metal liq.-like, surfactant effects on formation of)
ΤT
     Interfacial tension
        (of surfactant solns., silver metal liq.-like film
        formation in relation to)
ΙT
     Sulfonic acids, compounds
     RL: PRP (Properties)
        (perfluoroalkane, ammonium and potassium salts, surfactant
        effect of, on silver metal liq.-like film formation)
ΙT
     Surfactants
        (silver metal liq.-like film formation in presence of)
     Carboxylic acids, compounds
ΤТ
     RL: PRP (Properties)
        (perfluoro, ammonium salts, surfactant effect of, on
        silver metal liq.-like film formation)
     7440-22-4, Silver, uses and miscellaneous
     RL: USES (Uses)
        (liq.-like metal film formation by, surfactant effects on)
ΤТ
     577-11-7 2885-00-9, 1-Octadecanethiol 9002-93-1,
                  52584-45-9, Monflor 31 57534-41-5, Zonyl FSN
     Triton X 100
     60529-61-5, Monflor 32 67479-85-0, Zonyl FSC 67479-86-1, Zonyl FSP
     RL: PRP (Properties)
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(silver metal liq.-like film formation in presence of)

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L15 ANSWER 12 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
     1987:33631 CAPLUS
ΑN
    106:33631
DΝ
OREF 106:5655a,5658a
ED Entered STN: 07 Feb 1987
    Maleimide copolymer and thermoplastic resin prepared by using this
    Kimura, Atsushi; Toyooka, Yutaka; Kishida, Kazuo
ΙN
PΑ
    Mitsubishi Rayon Co., Ltd., Japan
SO
    PCT Int. Appl., 41 pp.
    CODEN: PIXXD2
DТ
    Patent
LA
    Japanese
IC
    ICM C08F002-18
     ICS C08F212-04; C08L033-14; C08L035-06; C08L051-04
CC
     35-4 (Chemistry of Synthetic High Polymers)
     Section cross-reference(s): 38
FAN.CNT 1
                                      APPLICATION NO. DATE
    PATENT NO.
                      KIND DATE
                       A1 19860731 WO 1986-JP17
    WO 8604337
PΤ
                                                               19860117
        W: AU, US
        RW: DE, FR, GB, IT, NL
    JP 61163903 A 19860724 JP 1985-4907
                                                                 19850117
     JP 61174248
                        A
                              19860805 JP 1985-12705
                                                                 19850128
    AU 8653567 A 19860813 AU 1986-53567
EP 208790 A1 19870121 EP 1986-900840
                                                                 19860117
                                                                 19860117
        R: DE, FR, GB, IT, NL
CA 1262299 A1 19891010 CA 1986-518902
PRAI JP 1985-4907 A 19850117
JP 1985-12705 A 19850128
WO 1986-JP17 A 19860117
                                                                 19860923
    WO 1986-JP17
                              19860117
                        A
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 WO 8604337
               ICM
                       C08F002-18
                       C08F212-04; C08L033-14; C08L035-06; C08L051-04
                IPCI
                       C08F0002-18 [ICM, 4]; C08F0002-12 [ICM, 4, C*];
                       C08F0212-04 [ICS, 4]; C08F0212-00 [ICS, 4, C*];
                       C08L0033-14 [ICS, 4]; C08L0033-00 [ICS, 4, C*];
                       C08L0035-06 [ICS, 4]; C08L0035-00 [ICS, 4, C*];
                       C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
                       C08F0002-12 [I,C*]; C08F0002-18 [I,A]; C08F0222-00
                IPCR
                       [I,C*]; C08F0222-40 [I,A]; C08L0035-00 [I,C*];
                       C08L0035-06 [I,A]; C08L0051-00 [I,C*]; C08L0051-04
                       [I,A]
                ECLA
                       C08F222/40; C08L035/06+B5; C08L051/04+B2
 JP 61163903
                IPCI
                       C08F0002-18 [ICM, 4]; C08F0002-12 [ICM, 4, C*];
                       C08F0212-04 [ICS, 4]; C08F0212-00 [ICS, 4, C*];
                       C08F0002-00 [ICA, 4]
 JP 61174248
                       C08L0033-18 [ICM, 4]; C08L0033-00 [ICM, 4, C*];
                IPCI
                       C08L0035-06 [ICS, 4]; C08L0035-00 [ICS, 4, C*];
                       C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
                       C08L0033-00 [I,C*]; C08L0033-00 [I,A]; C08L0007-00
                       [I,C*]; C08L0007-00 [I,A]; C08L0021-00 [I,C*];
                       C08L0021-00 [I,A]; C08L0023-00 [I,C*]; C08L0023-00
                       [I,A]; C08L0033-02 [I,A]; C08L0033-18 [I,A];
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C08L0033-24 [I,A]; C08L0035-00 [I,C*]; C08L0035-06
                        [I,A]; C08L0051-00 [I,C*]; C08L0051-00 [I,A];
                        C08L0051-02 [I,A]; C08L0051-04 [I,A]; C08L0101-00
                        [I,C*]; C08L0101-00 [I,A]
AU 8653567
                 IPCI
                        C08F0002-18 [ICM, 4]; C08F0002-12 [ICM, 4, C*];
                        C08F0212-04 [ICS, 4]; C08F0212-00 [ICS, 4, C*];
                        C08L0033-14 [ICS, 4]; C08L0033-00 [ICS, 4, C*];
                        C08L0035-06 [ICS, 4]; C08L0035-00 [ICS, 4, C*];
                        C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
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                        C08F0002-12 [I,C*]; C08F0002-18 [I,A]; C08F0222-00
                        [I,C*]; C08F0222-40 [I,A]; C08L0035-00 [I,C*];
                        C08L0035-06 [I,A]; C08L0051-00 [I,C*]; C08L0051-04
                        [I,A]
                        C08F222/40; C08L035/06+B5; C08L051/04+B2
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 EP 208790
                        C08F0002-18 [ICM, 4]; C08F0002-12 [ICM, 4, C*];
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                        C08L0033-14 [ICS, 4]; C08L0033-00 [ICS, 4, C*];
                        C08L0035-06 [ICS, 4]; C08L0035-00 [ICS, 4, C*];
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 CA 1262299
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                 IPCI
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                        C08F0212-00 [I,C*]; C08F0212-04 [I,A]; C08L0025-00
                 IPCR
                        [I,C*]; C08L0025-02 [I,A]; C08L0051-00 [I,C*];
                        C08L0051-04 [I,A]
    A maleimide polymer with excellent heat stability during high-temp.
AΒ
     molding and giving a product with excellent resistance to discoloration,
     heat, and impact when blended with a graft rubber, is prepd. by polymn. of
     a monomer selected from an arom. vinyl monomer, an unsatd. nitrile, and Me
     methacrylate 50-95, a maleimide 5-50, and other monomers 0-30\% in the
     presence of a Ca phosphate-based dispersing agent and a nonionic
     surfactant [RO(CH2CH2O)n]mPO(OA)3-m (R = C8-30 alkyl, aralkyl; A = H,
     metal; m = 1-3, n = 5-50). The process minimizes the scale formation of
     formed polymers on a reactor wall during polymn. Thus, a mixt. of
     arylonitrile 20, styrene 170, and N-phenylmaleimide 10 parts in 100 parts
     water contg. AIBN 0.1, tert-Bu benzoate 0.1, tert-dodecyl mercaptan 0.3,
     Gafac GB 520 0.003, and Ca3PO4 0.5 part was suspension-polymd. at
     80° for 3 h and at 120° for 2 h to give polymer beads
     (particle diam. 180 \mu, glass-transition temp. 125°). During the
     polymn., no scale formation was obsd. A blend of 55 parts maleimide
     copolymer and 45 parts graft polymer from polybutadiene 50, acrylonitrile
     15, and styrene 35 parts contg. Mg stearate 0.3, tris(nonylphenyl)
     phosphite 0.1, and Antage W 400 0.2 phr was injection-molded at
     280-290° to give a sample exhibiting yellowing index (at
     280°) 31, notched Izod impact strength 16.4 kg-cm/cm2, Rockwell
     hardness (R) 102, and Vicat softening point 108°, with no silver
     streak formation, compared with 44, 16, 101, and 104, with silver streak
     formation, when a maleimide copolymer prepd. in the presence of poly(vinyl
     alc.) as a dispersing agent was used.
    phenylmaleimide copolymer suspension polymn; acrylonitrile copolymer
     suspension polymn; styrene copolymer suspension polymn; calcium phosphate
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dispersant suspension polymn; polyethylene glycol lauryl ether phosphate;

nonionic phosphate surfactant suspension polymn; scale prevention suspension polymn dispersant; ABS blend maleimide copolymer molding; heat stability maleimide copolymer molding Plastics, molded ΤT RL: USES (Uses) (ABS polymer-maleimide-contg. polymers, heat- and impact-resistant, heat-stable) ΤТ Heat-resistant materials (maleimide-contg. polymers, heat stability improvement of) ΙT Dispersing agents (polyalkylene glycol phosphate-tricalcium phosphate, in suspension polymn. of maleimide-contq. monomer mixts., for scale formation prevention) Scale (coating) ΙT (prevention of, on reactor wall during suspension polymn. of maleimide-contg. monomer mixts., dispersing agents for) ΙT Polymerization (suspension, of maleimide-contg. monomer mixts., dispersing agents for, for scale formation prevention) ΙT 9003-56-9 RL: USES (Uses) (phenylmaleimide copolymer blends, heat-stable, resistant to discoloration, heat and impact) ΙT 31621-07-5P, Acrylonitrile-N-phenylmaleimide-styrene copolymer 94858-30-7P, Acrylonitrile- α -methylstyrene-N-phenylmaleimide-styrene 101482-57-9P, Acrylonitrilemethyl copolymer methacrylate-N-phenylmaleimide-styrene copolymer RL: PREP (Preparation) (prepn. of, by suspension polymn., dispersing agents for, for improved heat stability and scale prevention during polymn.) 51811-79-1, Gafac RE 610 ΤT RL: USES (Uses) (suspensing agents, Gafac RE 610, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.) 35604-29-6, Gafac GB 520 ΤТ RL: USES (Uses) (suspension agent, Gafac GB 520, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.) 7758-87-4, Tricalcium phosphate ΤТ RL: USES (Uses) (suspension agent, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.) RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD RE (1) Anon; JP 4983785 A (2) Anon; JP 5495689 A (3) Anon; JP 57125242 A CAPLUS (4) Anon; JP 57167341 A CAPLUS (5) Anon; JP 58129043 A CAPLUS (6) Anon; JP 58206657 A CAPLUS (7) Anon; JP 59184243 A CAPLUS

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L10
L11
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L13
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L14
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       1054666 NON
        307432 IONIC
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        13094 ZWITTERIONIC
        275066 SURFACTANT#
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            3 L11 AND L16
=> d 1-3 all
L17 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
     2008:619503 CAPLUS
AN
    148:545131
DN
ED
    Entered STN: 23 May 2008
     Detergent compositions in the form of microemulsions and use thereof in
    the treatment of alopecia
     Ben Alloum, Abdelkrim
ΙN
PA
    Morocco
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SO
    PCT Int. Appl., 63pp.
    CODEN: PIXXD2
DT
    Patent
LA
    French
CC
    62-4 (Essential Oils and Cosmetics)
FAN.CNT 1
    PATENT NO.
                      KIND DATE APPLICATION NO. DATE
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    WO 2008060130
WO 2008060130
                       A2 20080522 WO 2007-MA11
PΙ
                       A3 20081211
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            MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL,
            PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN,
            TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW
        RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
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            BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW,
            GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
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PRAI MA 2006-29458
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CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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WO 2008060130 IPCI A61Q0007-00 [I,A]; A61Q0019-10 [I,A]; A61P0017-14
                      [I,A]; A61K0008-37 [I,A]; A61K0008-06 [I,A];
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                       [I,C]; A61K0008-37 [I,A]; A61K0008-39 [I,A];
                       A61K0008-55 [I,A]; A61K0008-72 [I,C]; A61K0008-86
                       [I,A]; A61P0017-00 [I,C]; A61P0017-14 [I,A];
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                       [I,A]; A61K0008-72 [I,C]; A61K0008-86 [I,A];
                       A61P0017-00 [I,C]; A61P0017-14 [I,A]; A61Q0019-10
                       [I,C]; A61Q0019-10 [I,A]
                      A61K008/06C; A61K008/36C; A61K008/92C; A61Q007/00
                ECLA
OS
    MARPAT 148:545131
AB
    The invention relates to detergent compns. in the form of stable,
    transparent oil-in-water-type microemulsions which are prepd. in
    accordance with the invention and which take the form of a liq. or gel.
    The aforementioned compns. comprise water, electrolytes, fatty acids, a
    combination of fatty acid salt type surface-active agents and at least one
    nonionic surface-active agent, an oil preferably selected from oils
    contg. long-chain triglycerides and, if desired, other auxiliary agents,
    additives and active principles. Said compns. can be used, in particular,
    to clean and condition keratinous matter such as hair or skin. The
    inventive compns. can micro-emulsify sebum on contact. The invention also
    relates to a cosmetic method for the treatment of androgenic alopecia or
    the prevention of hair loss. A microemulsion contained sunflower oil
    3.26, free fatty acids 2.51, glycerin 2.04, copra oil fatty acids 19.84,
    Tergitol NP-9 16.32, sodium lactate 1.02, sodium chloride 0.65, disodium
    EDTA 0.16, and water 54.20%. Efficacy of the compn. in patient with
    alopecia is shown.
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Alcohols, biological studies
ΙT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (C9-11, ethoxylated; detergent compns. in form of microemulsions and
        use in treatment of alopecia)
     Fatty acids, biological studies
ΙT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (alkali metal salts; detergent compns. in form of microemulsions and
        use in treatment of alopecia)
ΙT
     Phenols, biological studies
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (alkyl; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
    Alcohols, biological studies
ΙT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (amino; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
ΙT
     Surfactants
        (anionic; detergent compns. in form of microemulsions and use
        in treatment of alopecia)
     Fats and Glyceridic oils, biological studies
ΙT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (avocado; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
    Fats and Glyceridic oils, biological studies
ΤT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (babassu; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
     Fats and Glyceridic oils, biological studies
ΙT
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (borage seed; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
ΤТ
     Surfactants
        (cationic; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
ΙT
    Alopecia
     Cosmetic microemulsions
     Detergents
     Electrolytes
     Human
       Surfactants
        (detergent compns. in form of microemulsions and use in treatment of
        alopecia)
ΙT
    Amines, biological studies
     Coconut oil
     Corn oil
     Fatty acids, biological studies
     Linseed oil
     Olive oil
     Palm oil
     Soybean oil
     Sunflower oil
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (detergent compns. in form of microemulsions and use in treatment of
        alopecia)
ΤТ
     Alcohols, biological studies
     RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (ethoxylated; detergent compns. in form of microemulsions and use in
        treatment of alopecia)
     Fats and Glyceridic oils, biological studies
ΙT
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RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)

(evening primrose; detergent compns. in form of microemulsions and use in treatment of alopecia)

- IT Alkali metal salts
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (fatty acid salts; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Amides, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (fatty; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Glycerides, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (long-chain; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Alopecia

(male pattern; detergent compns. in form of microemulsions and use in treatment of alopecia)

- IT Surfactants
 - (nonionic; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Alcohols, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (polyhydric; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Fatty acids, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (potassium salts; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Amino acids, biological studies
 - Carboxylic acids, biological studies

Fatty acids, biological studies

- RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (salts; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Fatty acids, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (sodium salts; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Fats and Glyceridic oils, biological studies
 - RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (vegetable; detergent compns. in form of microemulsions and use in treatment of alopecia)
- IT Surfactants
 - (**zwitterionic**; detergent compns. in form of microemulsions and use in treatment of alopecia)
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(detergent compns. in form of microemulsions and use in treatment of alopecia)

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L17 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
     2005:216634 CAPLUS
ΑN
DN
    142:284776
ED
    Entered STN: 11 Mar 2005
    Method and compositions for straightening hair using a reducing and an
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     Mueller, Burkhard; Schellin, Aaltje; Neubueser, Inge
ΙN
PΑ
     Hans Schwarzkopf & Henkel GmbH & Co. KG, Germany
SO
     PCT Int. Appl., 49 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    German
IC
     ICM A61K007-09
     ICS A61K007-075
CC
     62-3 (Essential Oils and Cosmetics)
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                                                                  DATE
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     DE 10338883 A1 20050324 DE 2003-10338883 20030823
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EP 1656184 A1 20060517 EP 2004-764143 20040814
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B1 20080326
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     JP 2007503379 T 20070222 JP 2006-523591
                                                                   20040814
AT 390175 T 20080415 AT 2004-764143
ES 2300812 T3 20080616 ES 2004-764143
US 20060150344 A1 20060713 US 2005-297707
HK 1086211 A1 20080815 HK 2006-106137

PRAI DE 2003-10338883 A 20030823
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AΒ
     The invention concerns a method and compns. for straightening hair by (i)
     applying an aq. soln. that contains a keratin-reducing agent (A); (ii)
     rinsing the reducing-agent contg. soln. after a time period; (iii) drying
     the hair; (iv) exposing the hair to straightening under heat treatment at
     120-220 °C by ironing; (v) applying an aq. soln. that contains an
     oxidn. agent (B); (vi) rinsing off the soln. after a time period elapsed.
     Both A and B solns. contain conditioning agents selected from the group of
     cationic polymers, quaternary ammonium compds., silicones and protein
     hydrolyzates. Thus a component A included (wt./wt.%): 1,2-propylene
     qlycol 2.00; cetyl/stearyl alc. (50-50% mixt.) 9.00; Lanette E 0.50; Brij
     35 P 0.50; Natrosol 250 HR 0.25; ammonia (25% aq.soln.) 5.00; Turpinal SL
     0.25; ammonium thioglycolate (71% aq.soln.) 18.00; ammonium bicarbonate
     4.00; Promois Silk 1000 1.00; Dow Corning 1403 fluid 0.50; perfume 1.00;
     water to 100. Component B contained (wt./wt.%): cetearyl alc. 4.00;
     Eumulgin B3 0.50; ammonia (25% aq.soln.) 0.80; dipicolinic acid 0.10;
     Turpinal SL 1.70; Rheocare CTH(E) 1.00; hydrogen peroxide (50% aq.soln.)
     4.00; water to 100.
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- ST hair straightening compn reducing oxidizing agent conditioner heat
- IT Polysiloxanes, biological studies

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) ([(aminoethyl)amino]propyl hydroxy, di-Me; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Surfactants

(amphoteric; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Surfactants

(anionic; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Polyelectrolytes

(cationic; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Hair preparations

(conditioners; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Cyclosiloxanes

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (di-Me; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Quaternary ammonium compounds, biological studies

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (ester group-contg.; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Fibroin

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (hydrolyzates; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Onium compounds

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (imidazolium compds., 2-(C9-19 and C9-19-unsatd. alkyl)-1-[(C10-20 and C10-20-unsatd. amido)ethyl]-4,5-dihydro-1-Me, Me sulfates, Rewoquat W 575PG; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Heat treatment

Oxidizing agents

Reducing agents

Viscosity

(method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Polysiloxanes, biological studies

Protein hydrolyzates

Quaternary ammonium compounds, biological studies

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Surfactants

(nonionic; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Hair preparations

(straighteners; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

IT Surfactants

(zwitterionic; method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

112-02-7, Dehyquart A-CA 2809-21-4, Turpinal SL 5421-46-5, Ammonium ΤТ 7651-02-7, Tegoamide S18 7722-84-1, Hydrogen thioglycolate peroxide, biological studies 8045-77-0, Lanette E 9002-92-0, Brij 35 P 9004-62-0, Natrosol 250 HR 9006-65-9, Dimethicone 16962-53-1D, Trimethyl ammonium, alkyl halogenide derivs. 17000-00-9D, Methylammonium, trialkyl halogenide derivs. 17000-01-0D, Dimethylammonium, dialkyl halogenide derivs. 17301-53-0, Genamin KDMP 26062-79-3, Merquat 100 26161-33-1, Rheocare CTH(E) 31692-79-2, 32208-04-1, Dehyquart F75 81859-24-7, Polymer JR 400 Dimethiconol 195868-36-1, Phenyltrimethicone 205537-77-5, Dow Corning 1403 473664-54-9, Salcare SC 96

RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (method and compns. for straightening hair using reducing and oxidizing agents in combination with heat)

RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD

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L17 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

- AN 2003:737150 CAPLUS
- DN 139:250305
- ED Entered STN: 19 Sep 2003
- TI Invisible patch for the controlled delivery of cosmetic, dermatological, and pharmaceutical active ingredients onto the skin
- IN Shefer, Adi; Shefer, Samuel
- PA USA
- SO U.S. Pat. Appl. Publ., 17 pp., Cont.-in-part of U.S. Ser. No. 91,935. CODEN: USXXCO
- DT Patent
- LA English
- IC ICM A61K031-715 ICS A61K009-70
- INCL 424449000; 514061000
- CC 63-6 (Pharmaceuticals)

Section cross-reference(s): 62

FAN.CNT 2

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ΡΙ	US CA WO	20030175333 20030175328 2515098			A1 A1 A1 A2		2003 2004 2004	0918 0916 0916	US 2003-376736 US 2002-91935 CA 2004-2515098 WO 2004-US6106						20020306 20040227					
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US 20030175333			_			51K031-715 51K009-70														

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      [I,C*]; A61F0013-00 [I,A]; A61K [I,S]; A61K0008-30
       [I,C]; A61K0008-33 [I,A]; A61K0008-34 [I,A];
      A61K0008-36 [I,A]; A61K0008-37 [I,A]; A61K0008-41
       [I,A]; A61K0008-43 [I,A]; A61K0008-46 [I,A];
      A61K0008-49 [I,A]; A61K0008-55 [I,A]; A61K0008-58
      [I,A]; A61K0008-60 [I,A]; A61K0008-72 [I,C];
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      [I,A]; A61K0008-88 [I,A]; A61K0009-70 [I,C];
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       [I,A]; A61K0047-10 [I,C]; A61K0047-10 [I,A];
      A61K0047-12 [I,C]; A61K0047-12 [I,A]; A61K0047-14
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       [I,A]; A61K0047-36 [I,C]; A61K0047-36 [I,A];
      A61K0047-38 [I,C]; A61K0047-38 [I,A]; A61K0047-42
      [I,C]; A61K0047-42 [I,A]; A61M0037-00 [I,C*];
      A61M0037-00 [I,A]; A61Q0019-00 [I,C]; A61Q0019-00 [I,A]
FTERM 4C076/AA72; 4C076/AA95; 4C076/BB31; 4C076/CC01;
       4C076/CC03; 4C076/CC04; 4C076/CC18; 4C076/DD03;
       4C076/DD04; 4C076/DD07; 4C076/DD08; 4C076/DD09;
       4C076/DD13; 4C076/DD17; 4C076/DD38A; 4C076/DD66A;
       4C076/EE06A; 4C076/EE10A; 4C076/EE12A; 4C076/EE13A;
       4C076/EE17A; 4C076/EE23A; 4C076/EE26A; 4C076/EE27;
       4C076/EE30A; 4C076/EE31A; 4C076/EE32A; 4C076/EE38A;
       4C076/FF31; 4C076/FF35; 4C083/AA112; 4C083/AB032;
       4C083/AC122; 4C083/AC131; 4C083/AC181; 4C083/AC371;
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       4C083/AC682; 4C083/AC772; 4C083/AC781; 4C083/AC791;
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       4C083/AD151; 4C083/AD201; 4C083/AD202; 4C083/AD211;
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       4C083/EE13; 4C083/EE14; 4C083/EE16; 4C083/EE22;
       4C084/AA17; 4C084/MA32; 4C084/MA63; 4C084/NA10;
       4C084/ZA891
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AB The present invention relates to a patch for controlled topical or transdermal delivery of effective levels of cosmetic, dermatol., and pharmaceutical active ingredients onto the skin, hair follicles, and sebaceous glands, with minimal discomfort and ease of use. The patch can be transparent or clear and comprises a rate-controlling matrix layer. The matrix layer comprises water-sensitive, bioadhesive, film forming polymers, a water sol. oligomer, and a surfactant. The cosmetic, dermatol., and pharmaceutical active ingredients are sol. or dispersed in the matrix. The patch becomes tacky when wetted and adheres onto the skin. The adhesive properties of the patch are sufficient to maintain the patch in place on the skin for the recommended treatment period while allowing the patch to be readily removed without causing skin irritation or leaving adhesive residue on the skin. For example, an antibiotic patch contained polyvinyl alc. 50, PVP 1, polysorbate 20 5, Maltrin 180 10, lactitol 5, glycerin 10, and chloramphenicol 0.55%. ST

patch bioadhesive polymer oligosaccharide **surfactant**; antibiotic patch PVA PVP polysorbate chloramphenicol

```
ΙT
     Glycosides
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (alkyl polyglycosides; invisible patches contg. bioadhesive polymers
        and surfactants)
ΙT
     Surfactants
        (amphoteric; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (anionic; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (cationic; invisible patches contq. bioadhesive polymers and
        surfactants)
     Essential oils
ΤТ
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (clove; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Hair preparations
        (conditioners; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
        (depilatories; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Acne
     Burn
     Dandruff
     Pruritus
     Rhus diversiloba
     Rhus toxicodendron
        (drugs for; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
    Alcohols, biological studies
     Amides, biological studies
     Esters, biological studies
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (ethoxylated; invisible patches contg. bioadhesive polymers and
        surfactants)
ΤТ
    Hair preparations
        (growth stimulants; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Vein, disease
        (hemorrhoid, drugs for; invisible patches contq. bioadhesive polymers
        and surfactants)
ΤТ
     Syrups (sweetening agents)
        (hydrolyzed starch; invisible patches contq. bioadhesive polymers and
        surfactants)
     Allergy inhibitors
     Aloe barbadensis
     Analgesics
     Anti-infective agents
     Anti-inflammatory agents
     Antibacterial agents
     Antibiotics
     Antiemetics
     Antihistamines
     Antimicrobial agents
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Antioxidants

```
Antiperspirants
     Antitussives
     Antiviral agents
     Chelating agents
     Chemotherapy
     Cholinergic antagonists
     Deodorants
     Disinfectants
     Fungicides
     Hemostatics
     Immunomodulators
     Insecticides
     Radical scavengers
     Sunscreens
     Suntanning agents
     Vasoconstrictors
     Vasodilators
     Wound healing promoters
        (invisible patches contg. bioadhesive polymers and surfactants
    Amine oxides
ΤТ
     Amino acids, biological studies
     Carbohydrates, biological studies
     Caseins, biological studies
     Flavonoids
     Gelatins, biological studies
     Glycerides, biological studies
     Lanolin
     Lecithins
     Oligosaccharides, biological studies
     Paraffin oils
     Peptides, biological studies
     Polyamides, biological studies
     Polyesters, biological studies
     Polyoxyalkylenes, biological studies
     Polyoxyalkylenes, biological studies
     Polysaccharides, biological studies
     Proteins
     Retinoids
     Vitamins
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
        )
ΤТ
     Anesthetics
        (local; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
    Cosmetics
        (moisturizers; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (nonionic; invisible patches contg. bioadhesive polymers and
        surfactants)
     Amines, biological studies
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polyamines, nonpolymeric; invisible patches contq. bioadhesive
        polymers and surfactants)
     Alcohols, biological studies
ΙT
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
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USES (Uses)
        (polyhydric, propoxylated; invisible patches contg. bioadhesive
       polymers and surfactants)
ΤТ
    Quaternary ammonium compounds, biological studies
    RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
    USES (Uses)
        (polymers; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
    Skin, disease
        (rash, drugs for; invisible patches contq. bioadhesive polymers and
       surfactants)
ΙT
    Cosmetics
        (skin-lightening; invisible patches contg. bioadhesive polymers and
        surfactants)
ΤT
    Drug delivery systems
        (tapes; invisible patches contg. bioadhesive polymers and
       surfactants)
ΙT
    Cosmetics
        (wrinkle-preventing; invisible patches contg. bioadhesive polymers and
       surfactants)
ΙT
    Surfactants
        (zwitterionic; invisible patches contg. bioadhesive polymers
        and surfactants)
    36574-66-0D, N-coco acyl derivs.
ΤT
    RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
    USES (Uses)
        (cocoamidopropylbetaine; invisible patches contq. bioadhesive polymers
        and surfactants)
    68-26-8, Retinol
                       96-26-4, Dihydroxyacetone
ΙT
                                                 814-71-1, Calcium
    thioglycolate
                    34452-51-2, Potassium thioglycolate
    RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
ΙT
    50-70-4, Sorbitol, biological studies 50-70-4D, Sorbitol, oligomers
             50-78-2, Aspirin 50-81-7, Vitamin C, biological studies
    50-99-7D, Glucose, esters 50-99-7D, D-Glucose, oligomers contg.
    55-56-1, Chlorhexidine 56-81-5, Glycerin, biological studies
                                                                    56-86-0D,
    Glutamic acid, N-acyl derivs. 57-48-7D, Fructose, oligomers contg.
    57-50-1D, Sucrose, esters 57-50-1D, Sucrose, oligomers contg. 57-55-6,
    Propylene glycol, biological studies 58-86-6D, Xylose, oligomers contg.
    59-23-4D, Galactose, oligomers contg. 59-87-0, Nitrofurazone 60-54-8,
    Tetracycline
                  69-65-8D, Mannitol, oligomers contg. 69-72-7, Salicylic
    acid, biological studies 69-79-4D, Maltose, oligomers contq.
                                                                    87-99-0D.
    Xylitol, oligomers contq. 106-11-6, Diethylene glycol monostearate
    107-36-8D, Isethionic acid, cocoyl derivs. 108-46-3, Resorcinol,
    biological studies 108-95-2, Phenol, biological studies 114-07-8,
                  115-83-3, Pentaerythritol tetrastearate 144-55-8, Sodium
    Ervthromycin
    bicarbonate, biological studies 151-21-3, Sodium lauryl sulfate,
    biological studies 404-86-4, Capsaicin 497-19-8, Sodium carbonate,
    biological studies 585-86-4D, Lactitol, oligomers contq. 585-88-6D,
    Maltitol, oligomers contg. 770-35-4, Phenoxyisopropanol 1338-41-6,
    Sorbitan monostearate 1406-18-4, Vitamin E
                                                   2216-51-5 3380-34-5,
    Triclosan 3458-28-4D, D-Mannose, oligomers contg.
                                                        6284-40-8
    7439-97-6, Mercury, biological studies 7440-22-4, Silver, biological
             7553-56-2, Iodine, biological studies 8011-96-9, Calamine
    studies
    8050-81-5, Simethicone 9000-01-5, Gum arabic 9002-89-5, Polyvinyl
    alcohol 9002-98-6 9003-05-8, Polyacrylamide 9003-39-8,
    Polyvinylpyrrolidone 9004-64-2, Hydroxypropyl cellulose 9005-25-8,
```

Starch, biological studies 9005-25-8D, Starch, hydrolyzates

Polysorbate 20 9011-13-6, Styrene-maleic anhydride copolymer

9005-64-5,

```
9011-16-9, Methyl vinyl ether-maleic anhydride copolymer 11099-07-3,
    Glycerin stearate 11111-12-9, Cephalosporin 11140-06-0, Glycerin
    palmitate 12694-22-3, Diglyceryl monostearate 13718-94-0D, Palatinose,
    oligomers contg. 15687-27-1, Ibuprofen 18323-44-9, Clindamycin
    30233-64-8, Glyceryl monobehenate 39529-26-5, Decaglyceryl decastearate
    42852-72-2 53998-08-6, Sarcosinate 63119-59-5, Diglycerin distearate
    68424-04-4, Polydextrose 71185-87-0, Hexaglyceryl tristearate
    75537-01-8, Gantrez S-97 95461-64-6, Decaglyceryl pentastearate
    99734-29-9, Tetraglyceryl tristearate 99880-64-5, Glyceryl dibehenate
    106392-12-5, Polyoxyethylene polyoxypropylene block copolymer
    RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
    USES (Uses)
       (invisible patches contg. bioadhesive polymers and surfactants
ΙT
    56-75-7, Chloramphenicol 94-09-7, Benzocaine
    RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
       (invisible patches contg. bioadhesive polymers and surfactants
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L9
L10
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L11
           319 S L9 AND L10
L12
          6680 S METAL TREAT##### OR TREAT###### METAL
L13
             1 S L11 AND L12
L14
        522045 S SILVER OR AG
L15
           12 S L11 AND L14
L16
          1173 S (NONIONIC OR NON-IONIC) AND ANIONIC AND ZWITTERIONIC AND SURF
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L18 ANSWER 1 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
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AN
    149:312963
DN
ED Entered STN: 29 Aug 2008
TI Preparation of conductive supported noble metal nanoparticle catalysts
IN Stucky, Galen D.; Zheng, Nanfeng
    The Regents of the University of California, USA
PΑ
SO
    U.S. Pat. Appl. Publ., 35pp.
    CODEN: USXXCO
    Patent
DT
   English
INCL 428403000; 216055000; 428402000; 264005000; 264007000; 5021000000;
    502300000; 502159000; 502355000; 502350000
    56-4 (Nonferrous Metals and Alloys)
    Section cross-reference(s): 57, 67
FAN.CNT 1
    PATENT NO.
                      KIND DATE
                                     APPLICATION NO. DATE
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    _____
                                         _____
                                                               ______
PI US 20080206562
                       A1 20080828
                                        US 2008-13436
                                                             20080112
PRAI US 2007-884668P
                       P
                             20070112
CLASS
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US 20080206562 INCL 428403000; 216055000; 428402000; 264005000; 264007000;
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                      [I,C*]; B01J0027-02 [I,A]; B01J0027-24 [I,A];
                      B01J0031-02 [I,A]; B01J0023-755 [I,A]; B01J0031-26
                       [I,A]
                NCL
                       428/403.000; 216/055.000; 216/083.000; 264/005.000;
                       264/007.000; 428/402.000; 502/080.000; 502/087.000;
                       502/100.000; 502/150.000; 502/159.000; 502/167.000;
                       502/168.000; 502/171.000; 502/180.000; 502/181.000;
                       502/200.000; 502/216.000; 502/232.000; 502/300.000;
                       502/325.000; 502/337.000; 502/339.000; 502/340.000;
                       502/344.000; 502/345.000; 502/347.000; 502/349.000;
                       502/350.000; 502/355.000
AΒ
    The prepn. of elec.-conductive noble metal nanoparticle catalysts on
    catalyst supports such as alumina, silica, titania, clays, zeolites, or
    carbon black, is described.
    gold silver palladium nanocatalyst support sol gel micelle ceramic
ST
ΤТ
    Solvents
```

(aprotic; prepn. of conductive supported noble metal nanoparticle

```
catalysts)
ΙT
     Polyethers, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (arom., alkyl-, surfactants; prepn. of conductive supported
        noble metal nanoparticle catalysts)
ΙT
     Thiols, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (caps on catalyst nanoparticles; prepn. of conductive supported noble
        metal nanoparticle catalysts)
     Bentonite, processes
ΙT
     Carbon black, processes
     Clays, processes
     Diatomite
     Silica gel, processes
     Zeolites (synthetic), processes
     RL: PEP (Physical, engineering or chemical process); TEM (Technical or
     engineered material use); PROC (Process); USES (Uses)
        (catalyst supports; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
    Nanoparticles
        (catalysts; prepn. of conductive supported noble metal nanoparticle
        catalysts)
    Alcohols, uses
ΤТ
     RL: MOA (Modifier or additive use); USES (Uses)
        (ethoxylated, surfactants; prepn. of conductive supported
        noble metal nanoparticle catalysts)
ΙT
     Hydrocarbons, processes
     RL: PEP (Physical, engineering or chemical process); TEM (Technical or
     engineered material use); PROC (Process); USES (Uses)
        (fluoro, catalyst supports; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΤТ
     Surfactants
        (in coatings; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΤТ
     Electroluminescent devices
     Molecular electronic devices
     Optoelectronics
     Secondary batteries
     Semiconductor devices
     Sensors
     Solar cells
        (nanocatalysts for; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     Photolysis catalysts
        (nanocatalysts; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΙT
     Catalysts
     Semiconductor materials
        (nanoparticles; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΙT
     Surfactants
        (nonionic; prepn. of conductive supported noble metal
        nanoparticle catalysts)
     Silsesquioxanes
ΤТ
     RL: RGT (Reagent); RACT (Reactant or reagent)
        (octyl- and hexyl-; prepn. of conductive supported noble metal
        nanoparticle catalysts)
ΙT
     Dyes
        (org.-, functional mol.; prepn. of conductive supported noble metal
```

nanoparticle catalysts)

```
Calcination
ΙT
    Catalyst supports
    Etching
    Reducing agents
        (prepn. of conductive supported noble metal nanoparticle catalysts)
ΙT
    7440-44-0, Carbon, processes
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
    engineered material use); PROC (Process); USES (Uses)
        (activated, catalyst supports; prepn. of conductive supported noble
       metal nanoparticle catalysts)
    64-17-5, Ethanol, uses
TΤ
    RL: NUU (Other use, unclassified); USES (Uses)
        (buffer soln.; prepn. of conductive supported noble metal nanoparticle
       catalysts)
    49543-63-7, 4-(tert-Butyl)benzyl mercaptan
ΤТ
    RL: MOA (Modifier or additive use); USES (Uses)
        (cap on nanoparticles; prepn. of conductive supported noble metal
       nanoparticle catalysts)
    1322-36-7, Dodecanethiol
ΙT
    RL: MOA (Modifier or additive use); USES (Uses)
        (caps on catalyst nanoparticles; prepn. of conductive supported noble
       metal nanoparticle catalysts)
    7440-02-0P, Nickel, preparation
                                    7440-05-3P, Palladium, preparation
    7440-06-4P, Platinum, preparation 7440-22-4P, Silver,
    preparation 7440-50-8P, Copper, preparation 7440-57-5P, Gold,
    preparation 12006-51-8P, AuCu
    RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation);
    USES (Uses)
        (catalyst nanoparticles; prepn. of conductive supported noble metal
       nanoparticle catalysts)
    1309-48-4, Magnesium oxide (MgO), processes 1314-23-4, Zirconia,
ΤТ
               1344-28-1, Aluminum oxide (Al2O3), processes 7631-86-9,
    processes
    Silica, processes 7782-42-5, Graphite, processes 13463-67-7, Titania,
    processes
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
    engineered material use); PROC (Process); USES (Uses)
        (catalyst supports; prepn. of conductive supported noble metal
       nanoparticle catalysts)
    1306-38-3, Cerium oxide (CeO2), uses
                                         1313-13-9, Manganese oxide (MnO2),
ΙT
          1313-96-8, Niobium oxide (Nb2O5)
    RL: MOA (Modifier or additive use); USES (Uses)
        (coatings on colloidal silica; prepn. of conductive supported noble
       metal nanoparticle catalysts)
ΙT
    12638-19-6P
    RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation);
    USES (Uses)
        (nanoparticles; prepn. of conductive supported noble metal nanoparticle
        catalysts)
ΤТ
    2966-50-9, Silver trifluoroacetate 14024-17-0, Iron acetyl
    acetonate 14024-61-4 14024-64-7 16902-59-3 17927-72-9
                                                                  19443-16-4
    19443-17-5
                23894-00-0
                            23894-03-3 24772-51-8
                                                      27858-32-8, Titanium
    diisopropoxide bis(ethyl acetoacetate) 62905-51-5 65574-21-2
    204522-78-1 299957-41-8 380240-62-0 1050499-47-2
                                                           1050499-48-3
    1050499 - 49 - 4 \qquad 1050499 - 50 - 7 \qquad 1050499 - 51 - 8 \qquad 1050499 - 52 - 9 \qquad 1050499 - 53 - 0
    1050499-54-1
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (precursors; prepn. of conductive supported noble metal nanoparticle
```

catalysts)

```
ΙT
    1313-99-1, Nickel oxide, uses 1345-25-1, Ferrous oxide, uses
    11104-61-3, Cobalt oxide
    RL: MOA (Modifier or additive use); USES (Uses)
        (prepn. of conductive supported noble metal nanoparticle catalysts)
    78-07-9, Ethyltriethoxysilane 78-10-4, Tetraethoxysilane 681-84-5, Tetramethoxysilane 682-01-9, Tetrapropoxysilane 1185-55-3,
ΤТ
    Methyltrimethoxysilane 1336-21-6, Ammonium hydroxide ((NH4)(OH))
    2031-67-6, Methyltriethoxysilane 4766-57-8, Tetrabutoxysilane
    30232-12-3 192082-40-9, Mercaptoundecanoic acid
    RL: RGT (Reagent); RACT (Reactant or reagent)
       (prepn. of conductive supported noble metal nanoparticle catalysts)
    1722-26-5, Triethylamine-borane 4856-95-5
                                               7337-45-3,
ΙT
    tert-Butylamine-borane 13774-81-7, Ammonia-borane
    RL: RGT (Reagent); RACT (Reactant or reagent)
       (reducing agents; prepn. of conductive supported noble metal
       nanoparticle catalysts)
ΙT
    67-66-3, Chloroform, uses 71-43-2, Benzene, uses 75-09-2,
    Dichloromethane, uses 108-88-3, Toluene, uses 110-54-3, Hexane, uses
    110-82-7, Cyclohexane, uses
    RL: NUU (Other use, unclassified); USES (Uses)
       (solvent; prepn. of conductive supported noble metal nanoparticle
       catalysts)
    14243-64-2
ΤТ
    RL: PEP (Physical, engineering or chemical process); PROC (Process)
       (substrates; prepn. of conductive supported noble metal nanoparticle
       catalysts)
    577-11-7, Sodium bis(2-ethylhexyl) sulfosuccinate 9002-89-5, Polyvinyl
ΙT
    alcohol 9002-92-0, Brij 30 9004-98-2, Brij 97 9036-19-5,
    (Octylphenoxy)polyethoxyethanol 12441-09-7D, Sorbitan, ester derivs.
    27251-32-7
    RL: MOA (Modifier or additive use); USES (Uses)
       (surfactants; prepn. of conductive supported noble metal
       nanoparticle catalysts)
L18 ANSWER 2 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
AN 2007:415891 CAPLUS
DN 146:463862
ED Entered STN: 16 Apr 2007
TI Discoloration prevention of metals using organic ultra-thin films and
    methods therefor
IN Liang, Chenghao; Yang, Changjiang; Huang, Naibao
PA Dalian Maritime University, Peop. Rep. China
SO
    Faming Zhuanli Shenging Gongkai Shuomingshu, 10pp.
    CODEN: CNXXEV
DT
    Patent
    Chinese
LA
    42-10 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 46, 56
FAN.CNT 1
    PATENT NO.
                 KIND DATE
                                   APPLICATION NO. DATE
                      ____
                              _____
                                         ______
                       A 20070411 CN 2006-10134093 20061026
PI CN 1943882
PRAI CN 2006-10134093
                             20061026
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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CN 1943882
              IPCI B05D0007-14 [I,A]; B05D0007-24 [I,A]; B05D0003-10
                      [I,A]; C23C0022-05 [I,A]; C07C0321-04 [I,A];
                      C07C0321-00 [I,C*]
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IPCR B05D0007-14 [I,C]; B05D0007-14 [I,A]

- OS MARPAT 146:463862
- AB Film-forming solns. contain 0.001-1 mol/L alkyl thiols and 0.001-1 mol/L surfactants. Thus, a coating soln. on **Ag** contained stearyl thiol 15, polyethylene glycol nonylphenyl ether 7, hexadecyltrimethylammonium bromide 2, Pluronic 64 7 g/L.
- ST metal discoloration prevention coating **surfactant** thiol; **silver** discoloration prevention coating **surfactant** thiol
- IT Surfactants

(anionic; coating materials contg. thiols and
surfactants for discoloration prevention of metals)

IT Surfactants

(cationic; coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

IT Discoloration prevention

(coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

IT Quaternary ammonium compounds, uses

Thiols, uses

RL: TEM (Technical or engineered material use); USES (Uses) (coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

IT Coating materials

(discoloration-resistant; coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

- IT 691397-13-4, Pluronic L 64
 - RL: TEM (Technical or engineered material use); USES (Uses) (Pluronic L 64; coating materials contg. thiols and **surfactants** for discoloration prevention of metals)
- TT 57-09-0, Hexadecyltrimethylammonium bromide **2885-00-9**, Stearylmercaptan 7440-22-4, **Silver**, uses 9016-45-9, Polyethylene glycol nonylphenyl ether RL: TEM (Technical or engineered material use); USES (Uses)

RL: TEM (Technical or engineered material use); USES (Uses) (coating materials contg. thiols and **surfactants** for discoloration prevention of metals)

L18 ANSWER 3 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

- AN 2007:150387 CAPLUS
- DN 146:236227
- ED Entered STN: 09 Feb 2007
- TI Conductive adhesive composition comprising pressure sensitive adhesive and electrolyte
- IN Menon, Vinod P.; Kumar, Kanta; Nelson, Carl T.; Rizzardi, Don A.
- PA 3M Innovative Properties Company, USA
- SO U.S. Pat. Appl. Publ., 20pp. CODEN: USXXCO
- DT Patent
- LA English
- INCL 600391000; 600392000; 252500000
- CC 63-7 (Pharmaceuticals)

FAN.CNT 1

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	AU	AU 2006278717						20070215 AU 2006-278717								2	20060801		
	CA	CA 2617273					A1 20070215 CA 2006-2617273							2	20060801				
	WO	WO 2007019115					A1 20070215					006-1	20060801						
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GE, GH, GM, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP,
             KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN,
            MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU,
             SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG,
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         RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,
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             GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
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     EP 1917318
                               20080507
                                          EP 2006-789019
                         Α1
                                                                  20060801
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             IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR
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    MX 2008001425
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                                          MX 2008-1425
                                                                  20080129
     KR 2008040689
                        Α
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                                          KR 2008-702725
                                                                  20080201
     CN 101238189
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                              20080806 CN 2006-80028822
                                                                  20080204
     IN 2008CN00571
                        A
                              20081128
                                          IN 2008-CN571
                                                                  20080204
PRAI US 2005-197216
WO 2006-US29794
                         Α
                              20050804
                        W
                               20060801
CLASS
PATENT NO.
               CLASS PATENT FAMILY CLASSIFICATION CODES
 US 20070032719 INCL
                       600391000; 600392000; 252500000
                IPCI
                       A61B0005-04 [I,A]; H01B0001-12 [I,A]; H01B0001-00 [I,A]
                 IPCR
                       A61B0005-04 [I,C]; A61B0005-04 [I,A]; H01B0001-00
                       [I,C]; H01B0001-00 [I,A]; H01B0001-12 [I,C];
                       H01B0001-12 [I,A]
                       600/391.000; 252/500.000; 600/392.000
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                 ECLA
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                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
 AU 2006278717
                 IPCI
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                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
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                       C09J133/06+B2; H01B001/20; K61B; M08L; M08L; M08L
 CA 2617273
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                       A61B0005-0408 [I,A]; A61B0018-14 [I,A]; A61K0050-00
                       [I,A]; A61N0001-04 [I,A]; C09J0009-02 [I,A];
                       C09J0009-00 [I,C*]; C09J0011-06 [I,A]; C09J0011-02
                        [I,C*]
                 IPCR
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]; A61B0005-0408
                        [I,C]; A61B0005-0408 [I,A]; A61B0018-14 [I,C];
                       A61B0018-14 [I,A]; A61K0050-00 [I,C]; A61K0050-00
                        [I,A]; A61N0001-04 [I,C]; A61N0001-04 [I,A];
                        C09J0011-02 [I,C]; C09J0011-06 [I,A]
 WO 2007019115
                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
                 IPCI
                 IPCR
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
                 ECLA
                       C09J009/02; A61B005/0408F; A61N001/04; C09J133/04+B4;
                       C09J133/06+B2; H01B001/20; K61B; M08L; M08L; M08L
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                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
                       C09J0009-00 [I,C]; C09J0009-02 [I,A]
                 IPCR
 JP 2009503235
                 IPCI
                       C09J0201-00 [I,A]; C09J0009-02 [I,A]; C09J0009-00
                        [I,C*]; C09J0004-02 [I,A]; A61L0024-00 [I,A];
                        A61N0001-04 [I,A]
                 FTERM 4C053/BB04; 4C053/BB06; 4C053/BB07; 4C053/BB23;
                        4C053/BB35; 4C053/BB36; 4C081/AA10; 4C081/AA12;
                        4C081/AC04; 4C081/BB03; 4C081/BB04; 4C081/CA061;
                        4C081/CA071; 4C081/CA081; 4C081/CA101; 4C081/CA16;
                        4C081/CA181; 4C081/CA211; 4C081/CA281; 4C081/CE07;
                        4C081/CE09; 4C081/CE10; 4C081/DA02; 4C081/DA12;
                        4C081/DB07; 4C081/DC03; 4C081/DC04; 4J040/FA041;
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4J040/FA081; 4J040/FA091; 4J040/FA101; 4J040/FA131;

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4J040/FA141; 4J040/FA161; 4J040/FA281; 4J040/FA291;
                        4J040/HB04; 4J040/HB10; 4J040/HB11; 4J040/HB14;
                        4J040/HC01; 4J040/HD02; 4J040/HD18; 4J040/HD23;
                        4J040/JA03; 4J040/JB09; 4J040/KA12; 4J040/KA13;
                        4J040/KA32; 4J040/KA38; 4J040/KA39; 4J040/MA14;
                        4J040/NA02
MX 2008001425
               IPCI
                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
 KR 2008040689 IPCI
                       C09J0009-02 [I,A]; C09J0009-00 [I,C*]
 CN 101238189 IPCI
 IN 2008CN00571 IPCI
                       C09J0009-02 [ICM, 7]; C09J0009-00 [ICM, 7, C*]
    MARPAT 146:236227
OS
    A conductive adhesive compn. is provided and articles that include the
AΒ
     adhesive compn. as a component thereof. The conductive adhesive compn.
     comprises: (a) pressure sensitive adhesive; (b) electrolyte comprising
     water sol. or water dispersible org. chloride; and (c) humectant. In some
     embodiments, the conductive adhesive compn. is a bicontinuous compn.
     comprising an aq. phase and an oil phase, and the bicontinuous compn. may
     be derived from a polymerizable microemulsion compn., the microemulsion
     compn. comprising: an aq. phase comprising one or more hydrophilic
     monomers or oligomers and/or one or more amphiphilic monomers or oligomers
     in water, the water-sol. or water-dispersible org. chloride, surfactant
     and humectant; and an oil phase comprising one or more hydrophobic
    monomers or oligomers. Biomedical articles such as biomedical electrodes,
    may incorporate the foregoing adhesive as a component. For example,
     adhesive precursor comprised of acrylic acid 15 g, 2-hydroxyethyl
     methacrylate 20 g, tetrakis(hydroxymethyl)phosphonium chloride 11 g,
     1,3-butylene glycol 25 g, glycerol 10 g, water 19 g, Irgacure 2959 0.55 g
     and polyethylene glycol diacrylate 0.15 g. The precursor was coated using
     a knife coater onto a release liner as substrate. The knife was set so
     that a 25 mil (0.64 mm) thick coating was obtained. Polymn. was induced
     in the coated microemulsion by exposure to UV radiation. A total dose of
     1800 mJ/cm2 was applied over approx. 7 min, forming a conductive,
     bicontinuous adhesive. This conductive adhesive had an excellent adhesion
    to human skin.
    polymer acrylate electrolyte chloride conductive adhesive
ST
    Polyurethanes, biological studies
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (acrylates; conductive adhesive compn. comprising pressure sensitive
        adhesive and electrolyte)
    Electric conductors
ΙT
        (adhesive; conductive adhesive compn. comprising pressure sensitive
        adhesive and electrolyte)
ΙT
     Fats and Glyceridic oils, biological studies
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (almond, amidopropalkonium chloride; conductive adhesive compn.
        comprising pressure sensitive adhesive and electrolyte)
ΙT
     Surfactants
        (anionic; conductive adhesive compn. comprising pressure
        sensitive adhesive and electrolyte)
ΙT
     Fats and Glyceridic oils, biological studies
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (avocado, amidopropalkonium chloride; conductive adhesive compn.
        comprising pressure sensitive adhesive and electrolyte)
ΙT
     Surfactants
        (cationic; conductive adhesive compn. comprising pressure sensitive
        adhesive and electrolyte)
ΙT
     Onium compounds
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RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(chloride; conductive adhesive compn. comprising pressure sensitive

STN Columbus adhesive and electrolyte) ΙT Quaternary ammonium compounds, biological studies RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (chlorides; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Fatty acids, biological studies RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (coco, trimethylammonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Chain transfer agents Crosslinking agents Electrodes Electrolytes Human Humectants Hydrogels Surfactants (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Alcohols, uses Thiols, uses RL: NUU (Other use, unclassified); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) Acrylic polymers, biological studies ΤT RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) Sulfonium compounds ΙT RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Adhesives (conductive; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Sovbean oil RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (dimethylammonium chloride; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Surfactants (nonionic; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Chlorides, biological studies RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (org.; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT Adhesives (pressure-sensitive; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) Fatty acids, biological studies ΤT RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (tallow, bishydroxyethyl/dime quaternary ammonium compds.; conductive adhesive compn. comprising pressure sensitive adhesive and electrolyte) ΙT 558-13-4, Carbon tetrabromide 25103-09-7, Isooctyl thioglycolate

electrolyte)
IT 1070-70-8, 1,4-Butanediol diacrylate 1321-74-0, Divinylbenzene, reactions 10526-04-2, 1,8-Octanediol diacrylate 13048-33-4,

(conductive adhesive compn. comprising pressure sensitive adhesive and

RL: NUU (Other use, unclassified); USES (Uses)

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STN Columbus
     1,6-Hexanediol diacrylate
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (conductive adhesive compn. comprising pressure sensitive adhesive and
        electrolyte)
ΙT
     56-34-8, Tetraethylammonium chloride 56-37-1, Benzyltriethylammonium
     chloride 56-81-5, Glycerin, biological studies 56-93-9,
     Benzyltrimethylammonium chloride 57-55-6, Propylene glycol, biological
     studies 67-48-1 77-99-6, Trimethylolpropane 88-12-0D, polymer
     107-21-1, Ethylene glycol, biological studies 107-88-0, 1,3-Butanediol
     110-63-4, 1,4-Butanediol, biological studies 112-00-5,
     Dodecyltrimethylammonium chloride 112-02-7, Hexadecyltrimethylammonium
     chloride 112-03-8, Octadecyltrimethylammonium chloride 124-64-1,
     Tetrakis(hydroxymethyl)phosphonium chloride 139-08-2,
     Tetradecyldimethylbenzylammonium chloride 593-81-7D, Trimethylammonium
     chloride, coco fatty acid derivs. 7173-51-5 9004-98-2, Brij 98
     9042-76-6 17301-53-0, Behenyltrimethylammonium chloride 24567-53-1,
     Phosphonium chloride 25265-71-8, Dipropylene glycol 26570-48-9,
     Polyethylene oxide diacrylate 26597-36-4 32862-91-2, Oxonium chloride
     60182-11-8, Polyethylene glycol acrylate 93507-51-8 106797-53-9, IRGACURE 2959 123776-56-7 145687-02-1, Pemulen TR 2 463965-14-2
     923929-97-9 923929-99-1 924299-17-2, Hetoxol OL 35
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (conductive adhesive compn. comprising pressure sensitive adhesive and
        electrolyte)
     7783-90-6, Silver chloride, biological studies
ΙT
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (conductive ink soln.; conductive adhesive compn. comprising pressure
        sensitive adhesive and electrolyte)
L18 ANSWER 4 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
    2005:1062684 CAPLUS
ΑN
DN
    143:351549
ED
    Entered STN: 05 Oct 2005
```

- TI Water-based sulfur-containing composition chemical mechanical polishing of nonferrous metals
- IN Johns, Peter Gamon; Harrison, Clare Elizabeth
- PA Middlesex Silver Co. Limited, UK
- SO Brit. UK Pat. Appl., 29 pp. CODEN: BAXXDU

DT Patent

LA English

IC ICM C23F011-16 ICS C23F011-00

CC 57-7 (Ceramics)

Section cross-reference(s): 56

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PI			12666					2005			 GB 2	004-	7163	2	20040330					
	AU	GB 2412666 AU 2005229275						2008 2005	1013		AU 2			_	20050324					
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     EP 1730325
                         A1
                                20061213
                                           EP 2005-718135
                                                                   20050324
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     CN 1946878
                  A
                              20070411 CN 2005-80013434
                        A 20071220 JP 2007-505641 20050324

A 20070713 IN 2006-DN5356 20060915

A 20061116 MX 2006-10964 20060925

A1 20071206 US 2007-594477 20070702

A 20040330
     JP 2007537354
     IN 2006DN05356
     MX 2006010964
US 20070277906
PRAI GB 2004-7163
     WO 2005-GB50043 W
                               20050324
CLASS
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 GB 2412666
                       C23F011-16
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                       C23F011-00
                 IPCI C23F0011-10 [I,C]; C23F0011-16 [I,A]; C23F0011-00
                       [I,C]; C23F0011-00 [I,A]
                        C09G0001-00 [I,C*]; C09G0001-02 [I,A]; C11D0003-00
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                        C11D0003-34 [I,A]; C11D0011-00 [I,C*]; C11D0011-00
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                        C23F011/16; C23F011/16B
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                        C11D0003-00 [I,C*]; C09G0001-00 [I,C*]; C11D0003-34
 AU 2005229275
                        [I,C*]; C11D0011-00 [I,C*]; C23F0011-10 [I,C*];
                        C11D0003-00 [I,A]; C09G0001-02 [I,A]; C11D0003-34
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                        C11D0003-00 [I,C*]; C11D0003-00 [I,A]; C09G0001-00
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                        [I,C*]; C09G0001-02 [I,A]; C11D0003-34 [I,C*];
                        C11D0003-34 [I,A]; C11D0011-00 [I,C*]; C11D0011-00
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                        C11D0003-00 [I,A]; C11D0003-34 [I,C]; C11D0003-34
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                 ECLA
                        C23F011/16; C23F011/16B
 WO 2005095675
                 IPCI
                        C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*];
                        C11D0003-00 [ICS, 7]; C11D0003-34 [ICS, 7]; C11D0011-00
                        [ICS, 7]; C09G0001-02 [ICS, 7]; C09G0001-00 [ICS, 7, C*]
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                        C09G0001-00 [I,C*]; C09G0001-02 [I,A]; C11D0003-00
                        [I,C*]; C11D0003-00 [I,A]; C11D0003-34 [I,C*];
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                        [I,A]; C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                        C23F011/16; C23F011/16B
                 ECLA
 EP 1730325
                        C23F0011-16 [I,A]; C23F0011-10 [I,C*]; C11D0003-00
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                        [I,A]; C11D0003-34 [I,A]; C11D0011-00 [I,A];
                        C09G0001-02 [I,A]; C09G0001-00 [I,C*]
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                        C23F0011-10 [I,C]; C23F0011-16 [I,A]; C09G0001-00
                        [I,C]; C09G0001-02 [I,A]; C11D0003-00 [I,C];
                        C11D0003-00 [I,A]; C11D0003-34 [I,C]; C11D0003-34
                        [I,A]; C11D0011-00 [I,C]; C11D0011-00 [I,A]
                 ECLA
                        C23F011/16; C23F011/16B
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CN 1946878
                IPCI
                        C23F0011-16 [I,A]; C23F0011-10 [I,C*]; C11D0003-00
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                        C09G0001-02 [I,A]; C09G0001-00 [I,C*]
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                        C23F0011-10 [I,C]; C23F0011-16 [I,A]; C09G0001-00
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                        C11D0003-00 [I,A]; C11D0003-34 [I,C*]; C11D0003-34
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                        C23F011/16; C23F011/16B
                 IPCI
 JP 2007537354
                        C23C0022-58 [I,A]; C11D0003-34 [I,A]; C23C0022-68
                        [I,A]; C23C0022-05 [I,C*]; C11D0003-20 [I,A];
                        C11D0001-52 [I,A]; C11D0001-38 [I,C*]; C11D0001-72
                        [I,A]; C11D0001-79 [I,A]; C11D0001-755 [I,A];
                        C11D0001-75 [I,A]; C11D0001-722 [I,A]; C11D0001-14
                        [I,A]; C11D0001-02 [I,C*]; C11D0001-90 [I,A];
                        C11D0001-88 [I,C*]; C11D0003-04 [I,A]; C11D0001-68
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                        C23C0022-05 [I,C]; C23C0022-58 [I,A]; C09G0001-00
                        [I,C*]; C09G0001-02 [I,A]; C09K0003-14 [I,C];
                        C09K0003-14 [I,A]; C11D0001-02 [I,C]; C11D0001-14
                        [I,A]; C11D0001-38 [I,C]; C11D0001-52 [I,A];
                        C11D0001-68 [I,C]; C11D0001-68 [I,A]; C11D0001-72
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                        C11D0001-722 [I,A]; C11D0001-75 [I,C]; C11D0001-75
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                        C11D0001-79 [I,C]; C11D0001-79 [I,A]; C11D0001-88
                        [I,C]; C11D0001-90 [I,A]; C11D0003-00 [I,C*];
                        C11D0003-00 [I,A]; C11D0003-04 [I,C]; C11D0003-04
                        [I,A]; C11D0003-20 [I,C]; C11D0003-20 [I,A];
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                 FTERM 4H003/AB27; 4H003/AC02; 4H003/AC10; 4H003/AC13;
                        4H003/AD04; 4H003/BA12; 4H003/DA15; 4H003/EA12;
                        4H003/EA19; 4H003/EB05; 4H003/EB18; 4H003/EB21;
                        4H003/ED02; 4H003/FA05; 4K026/AA01; 4K026/AA06;
                        4K026/CA15; 4K026/CA37; 4K026/DA02; 4K026/DA03
 IN 2006DN05356 IPCI
                        C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                        C09G0001-02 [ICM, 7]; C09G0001-00 [ICM, 7, C*];
 MX 2006010964
                 IPCI
                        C11D0011-00 [ICS, 7]; C11D0003-00 [ICS, 7]; C11D0003-34
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                        C23F0011-16 [I,A]; C23F0011-10 [I,C*]; C09G0001-02
 US 20070277906 IPCI
                        [I,A]; C09G0001-00 [I,C*]; C11D0011-00 [I,A];
                        C11D0003-00 [I,A]; C11D0003-34 [I,A]
                 NCL
                        148/022.000
OS
    MARPAT 143:351549
AΒ
    A compn. and assocd. method of manuf. of a water based compn. comprising a
     treatment agent selected from an alkanethiol, alkyl thioglycollate, and
     dialkyl sulfide or dialkyl disulfide. The compn. also includes at least
     one of an amphoteric, non-ionic or cationic surfactant, where the
     treatment agent is directly dissolved or dispersed the water contq. the
     amphoteric, non-ionic or cationic surfactant. The compn. is
     particularly useful for the treatment of Ag-Cu-Ge alloy, copper, brass,
     and nickel. A solid polishing medium can also be included in the compn.,
     for example, silica or pptd. chalk, alumina, or silica.
ST
     chalk alumina silica alkanethiol thioglycollate chem mech polishing copper
     Thiols, processes
IT
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (alkanethiol; water-based sulfur-contq. compn. chem. mech. polishing of
```

```
metals)
ΙT
     Disulfides
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
        (alkyl; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΤТ
     Chalk
     Diatomite
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΙT
     Surfactants
        (cationic; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΤТ
     Polishing
        (chem.-mech.; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
ΙT
     Polishing materials
        (paste; water-based sulfur-contq. compn. chem. mech. polishing of
        metals)
ΤT
     Thioethers
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (water-based sulfur-contq. compn. chem. mech. polishing of metals)
ΙT
     1344-28-1, Alumina, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
     9004-82-4, Sodium laureth sulfate
ΤТ
     RL: MOA (Modifier or additive use); USES (Uses)
        (anionic surfactant; water-based sulfur-contg.
        compn. chem. mech. polishing of metals)
ΤТ
     7631-86-9, Silica, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-contg. compn. chem. mech. polishing of
        metals)
     36574-66-0D, N-coco acyl derivs.
TΤ
     RL: MOA (Modifier or additive use); USES (Uses)
        (cocamidopropyl betaine, surfactant; water-based
        sulfur-contg. compn. chem. mech. polishing of metals)
ΙT
     7440-02-0, Nickel, processes 7440-50-8, Copper, processes
     12597-71-6, Brass, processes
                                   74969-69-0
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); PROC (Process)
        (polished substrate; water-based sulfur-contg. compn. chem. mech.
        polishing of metals)
ΤТ
     62-56-6, Thiourea, uses 2885-00-9, Octadecyl mercaptan
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing compn. component; water-based sulfur-contg. compn. chem.
        mech. polishing of metals)
IT 2917-26-2, Hexadecyl mercaptan
     RL: MOA (Modifier or additive use); USES (Uses)
        (surfactant; water-based sulfur-contg. compn. chem. mech.
        polishing of metals)
     68-11-1D, alkyl esters
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
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(water-based sulfur-contg. compn. chem. mech. polishing of metals)
RE.CNT 6
            THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
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(2) Anon; GB 0956927 A
(3) Anon; GB 1117510 A
(4) Anon; US 3503883 A
(5) Anon; US 3518098 A
(6) Anon; US 5650385 A CAPLUS
L18 ANSWER 5 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
     2005:622423 CAPLUS
AN
    143:295501
DN
    Entered STN: 19 Jul 2005
ED
ΤI
     Single Etch Patterning of Stacked Silver and Molybdenum Alloy Layers on
    Glass Using Microcontact Wave Printing
ΑU
     Burdinski, Dirk; Brans, Harold J. A.; Decre, Michel M. J.
    Philips Research, Eindhoven, 5656 AA, Neth.
CS
    Journal of the American Chemical Society (2005), 127(31), 10786-10787
SO
    CODEN: JACSAT; ISSN: 0002-7863
PΒ
    American Chemical Society
DT
    Journal
LA
    English
CC
    74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 56
AΒ
    Stacked thin layers of silver alloy (AgPdCu) and MoCr layers on 10
     × 15 cm2 glass substrates were patterned by microcontact wave
     printing and etching. Patterns of etch-resistant octadecanethiol
     self-assembled monolayers (SAMs) were wave printed with regular backplane
     stabilized PDMS stamps. Pattern development was achieved by etching both
     metal layers in a single step, employing a nitric acid-based etching bath.
     Trifluoroacetic acid and a nitrite salt were identified as essential bath
     components for a homogeneous etching process. Etch defects could be
     eliminated by the addn. of a decanesulfonate, which stabilizes the SAM
     resist via a defect healing mechanism.
ST
     etching silver molybdenum alloy electrode display
ΙT
    Liquid crystal displays
        (active matrix; single etch patterning of stacked silver and
       molybdenum alloy layers on glass using microcontact wave printing as
       electrodes for)
     Surfactants
ΙT
        (anionic; single etch patterning of stacked silver
        and molybdenum alloy layers on glass using microcontact wave printing)
ΙT
     Lithography
        (microcontact printing; single etch patterning of stacked
        silver and molybdenum alloy layers on glass using microcontact
       wave printing)
    Autocatalysis
ΤT
     Electrodes
     Etching
     Glass substrates
     Self-assembled monolayers
        (single etch patterning of stacked silver and molybdenum
        alloy layers on glass using microcontact wave printing)
     64-19-7, Acetic acid, processes 76-05-1, Trifluoroacetic acid, processes
ΙT
                                7664-38-2, Phosphoric acid, processes
     7632-00-0, Sodium nitrite
     7697-37-2, Nitric acid, processes
```

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process) (etchant; single etch patterning of stacked silver and molybdenum alloy layers on glass using microcontact wave printing) IT 2885-00-9, 1-Octadecanethiol RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (ink, self-assembled monolayer; single etch patterning of stacked silver and molybdenum alloy layers on glass using microcontact wave printing) 188820-19-1 ΙT 317855-00-8 RL: CPS (Chemical process); DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (single etch patterning of stacked silver and molybdenum alloy layers on glass using microcontact wave printing) ΙT 13419-61-9, Sodium decane sulfonate RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process) (surfactant for etching soln.; single etch patterning of stacked **silver** and molybdenum alloy layers on glass using microcontact wave printing) THERE ARE 31 CITED REFERENCES AVAILABLE FOR THIS RECORD RE (1) Addison, C; Chem Rev 1980, V80, P21 CAPLUS (2) Balbaud, F; Corros Sci 2000, V42, P1685 CAPLUS (3) Beake, B; J Chem Soc, Perkin Trans 2 1998, P1 CAPLUS (4) Becka, A; J Phys Chem 1993, V97, P6233 CAPLUS (5) Boughriet, A; J Electroanal Chem 1986, V200, P217 CAPLUS (6) Creager, S; Langmuir 1993, V9, P2330 CAPLUS (7) Decre, M; Mater Res Soc Symp Proc 2004, VEXS-2, P59 CAPLUS (8) Delamarche, E; Langmuir 2003, V19, P5892 CAPLUS (9) Delamarche, E; Langmuir 2003, V19, P6567 CAPLUS (10) Francisco, J; J Chem Phys 2001, V115, P2117 CAPLUS (11) French, M; Langmuir 1998, V14, P2129 CAPLUS (12) Gates, B; Annu Rev Mater Res 2004, V34, P339 CAPLUS (13) Geissler, M; Adv Mater 2004, V16, P1249 CAPLUS (14) Geissler, M; Langmuir 2002, V18, P2374 CAPLUS (15) Geissler, M; Langmuir 2003, V19, P6283 CAPLUS (16) Goetting, L; Langmuir 1999, V15, P1182 CAPLUS (17) Hsieh, H; J Electrochem Soc 1992, V139, P1897 CAPLUS (18) Hsieh, H; J Electrochem Soc 1992, V139, P380 CAPLUS (19) Jackman, R; Science 1995, V269, P664 CAPLUS (20) Kumar, A; Appl Phys Lett 1993, V63, P2002 CAPLUS (21) Porter, M; J Am Chem Soc 1987, V109, P3559 CAPLUS (22) Rogers, J; Proc Natl Acad Sci U S A 2001, V98, P4835 CAPLUS (23) Schellekens, J; Mater Res Soc Symp Proc 2004, VEXS-2, P21 CAPLUS (24) Schmid, G; Ber Bunsen-Ges Phys Chem 1972, V76, P151 CAPLUS (25) Smith, R; Prog Surf Sci 2004, V75, P1 CAPLUS (26) Spitzer, U; J Org Chem 1974, V39, P3936 CAPLUS (27) Tate, J; Langmuir 2000, V16, P6054 CAPLUS (28) Tsujimura, T; J Vac Sci Technol, B 2002, V20, P1907 CAPLUS (29) Turney, T; Chem Rev 1959, V59, P497 (30) Ward, R; J Phys Chem B 1997, V101, P1594 CAPLUS (31) Xia, Y; J Electrochem Soc 1996, V143, P1070 CAPLUS L18 ANSWER 6 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN Full Text

2004:847649 CAPLUS

141:353637

AN DN

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Entered STN: 15 Oct 2004
ED
ΤI
     Pretreatment of Ag-alloy surface with organosulfur compounds for
     tarnishing prevention
     Johns, Peter Gammon; Harrison, Clare Elizabeth
IN
     Middlesex Silver Co. Limited, UK
PA
     PCT Int. Appl., 43 pp.
     CODEN: PIXXD2
     Patent
DT
     English
LA
IC
     ICM C23F011-16
CC
     56-6 (Nonferrous Metals and Alloys)
FAN.CNT 1
                                       APPLICATION NO. DATE
    WO 2004087996 A1 ^^^
     PATENT NO.
                        KIND DATE
                                           _____
                        A1 20041014 WO 2004-GB1373 20040330
PΤ
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
             CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
             GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
             LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,
             NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
             TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
         RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
             BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE,
             ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI,
             SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
             TD, TG
     AU 2004225693
                         A1
                              20041014 AU 2004-225693
                                                                   20040330
     CA 2520807
                        A1 20041014 CA 2004-2520807
A1 20060104 EP 2004-724313
                                                                  20040330
     EP 1611267
                                                                   20040330
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK
CN 1780937 A 20060531 CN 2004-80011375 20040330
JP 2006523266 T 20061012 JP 2006-506057 20040330
IN 2005DN04346 A 20070831 IN 2005-DN4346 20050926
MX 2005010452 A 20060510 MX 2005-10452 20050928
US 20070039665 A1 20070222 US 2005-551476 20050929
PRAI GB 2003-7290 A 20030331
WO 2004-GB1373 W 20040330
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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 WO 2004087996 ICM C23F011-16
                 IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                 IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                 ECLA C23F011/16; C23F011/16B
                      C23F0011-16 [ICM,7]; C23F0011-10 [ICM,7,C*]
 AU 2004225693
                 IPCI
                 IPCR
                      C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                 ECLA C23F011/16; C23F011/16B
 CA 2520807
                 IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                 IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                 ECLA C23F011/16; C23F011/16B
 EP 1611267
                 IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                 IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                       C23F011/16; C23F011/16B
                 ECLA
                 IPCI
                       C23F0011-16 [I,A]; C23F0011-10 [I,C*]
 CN 1780937
                 ECLA
                       C23F011/16; C23F011/16B
 IPCR
                       C23F0011-00 [I,C]; C23F0011-00 [I,A]; C22C0005-06
                        [I,C]; C22C0005-06 [I,A]; C22C0005-08 [I,A];
                        C23F0011-10 [I,C*]; C23F0011-16 [I,A]
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FTERM 4K062/AA01; 4K062/BB21; 4K062/BC22; 4K062/FA16
 IN 2005DN04346 IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
MX 2005010452
                IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                ECLA C23F011/16; C23F011/16B
US 20070039665 IPCI C23G0001-00 [I,A]; C23C0022-58 [I,A]; C23C0022-05
                        [I,C*]
                NCL
                        148/271.000; 134/002.000
AΒ
     The Ag alloys contq. minor Ge (esp. Ag-Cu-Ge alloys) to decrease the
     fire stain discoloration are pretreated on the surface with an
     alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide
     to prevent tarnishing. The treatment with organosulfur compds. is
     suitable for manufd. Ag-alloy articles to prevent tarnished appearance
     during transit and the subsequent extended display without special
     packaging. The Ag-alloy surface is optionally treated with aq. soln.
     contg. an alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl
     disulfide, as well as a mixt. of anionic surfactant and amphoteric or
     nonionic surfactant to solubilize the treatment agent. The typical
    ternary alloy contains Ag 80-96, Cu 1-19.9, and Ge 0.1-5%.
     silver copper germanium alloy tarnishing prevention organosulfur
ST
ΙT
     Surfactants
        (anionic, in tarnishing prevention; Ag-alloy
        surface treated with organosulfur compds. for tarnishing prevention)
ΙT
        (in tarnishing prevention; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
ΙT
     Surfactants
        (nonionic, in tarnishing prevention; Ag-alloy
        surface treated with organosulfur compds. for tarnishing prevention)
ΙT
     Tarnishing
        (prevention of; Aq-alloy surface treated with organosulfur
        compds. for tarnishing prevention)
     Thioethers
ΤT
     Thiols, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (tarnishing prevention by; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
    7440-56-4, Germanium, uses
ΤТ
     RL: MOA (Modifier or additive use); USES (Uses)
        (Ag alloys contg., tarnishing prevention on; Ag
        -alloy surface treated with organosulfur compds. for tarnishing
       prevention)
     106-94-5, n-Propyl bromide
ΤT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (solvent, in tarnishing prevention; Ag-alloy surface treated
        with organosulfur compds. for tarnishing prevention)
IT 2885-00-9, Octadecyl mercaptan 2917-26-2,
     Cetyl mercaptan
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnishing prevention by; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
ΤТ
     39282-03-6, Sterling silver
                                 103221-24-5 476614-10-5
     476614-12-7
                  476614-13-8
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnishing prevention on; Ag-alloy surface treated with
        organosulfur compds. for tarnishing prevention)
ΙT
     9080-17-5, Ammonium polysulfide
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
```

(test soln. with, for tarnishing; **Ag**-alloy surface treated with organosulfur compds. for tarnishing prevention)

RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Carlton, C; US 3503883 A 1970
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- (6) Goddard & Sons Ltd J; GB 1130540 A 1968
- (7) Han, S; JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 2001, V123, P2422 CAPLUS
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- L18 ANSWER 7 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

- AN 2003:851241 CAPLUS
- DN 139:330251
- ED Entered STN: 30 Oct 2003
- TI **Silver** (carboxylate-n-alkyl thiolate) particles for photothermographic of thermographic imaging
- IN Ghyzel, Peter J.; Lelental, Mark; Dickinson, David A.; Pitt, Alan R.;
 Wear, Trevor J.
- PA Eastman Kodak Company, USA
- SO U.S., 14 pp. CODEN: USXXAM
- CODEN: USXXA
- DT Patent
- LA English
- IC ICM G03C001-498
- INCL 430619000; 430611000; 430620000; 430631000
- CC 74-2 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

IAN.CIVI I																		
	PATENT NO.				KIND		DATE		AF	APPLICATION NO.					DATE			
							_											
ΡI	US	6638	708			В1		2003	1028	US	2	002-2	2004	17		20	0020	722
	EP 1385047			A1 20040128			EP 2003-77179						20030710					
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB, G	GR,	ΙΤ,	LI,	LU,	NL,	SE,	MC,	PT,
			ΙE,	SI,	LT,	LV,	FI,	, RO,	MK,	CY, A	L,	TR,	BG,	CZ,	EE,	HU,	SK	
	JP	2004	0542	76		A		2004	0219	JF	2	003-1	19929	97		20	030	718
PRAI	US	2002	-200	417		A		2002	0722									
CLASS																		
PAT	ENT	NO.		CLA	SS	PATE	I TV	FAMIL	Y CL	ASSIFI	CA	NOIT	CODI	ES				

PATENT NO. CLA		PATENT FAMILY CLASSIFICATION CODES						
US 6638708	ICM INCL IPCI	G03C001-498 430619000; 430611000; 430620000; 430631000 G03C0001-498 [ICM,7]						
	IPCR	B41M0005-30 [I,C*]; B41M0005-323 [I,A]; G03C0001-498 [I,C*]; G03C0001-498 [I,A]						
	NCL	430/619.000; 430/611.000; 430/620.000; 430/631.000						
	ECLA	G03C001/498B; G03C001/498E1						
EP 1385047	IPCI	G03C0001-498 [ICM, 7]						
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	ECLA	G03C001/498B; G03C001/498E1						
JP 2004054276	IPCI	G03C0001-498 [ICM,7]; B41M0005-30 [ICS,7]						
	IPCR	G03C0001-498 [I,A]; G03C0001-498 [I,C*]						
	FTERM	2H026/AA07; 2H026/BB46; 2H123/AB00; 2H123/AB03; 2H123/AB25; 2H123/AB28; 2H123/BC00; 2H123/BC12;						

2H123/CB00; 2H123/CB03

- AB The present disclosure relates to dispersions of **silver** (carboxylate-n-alkyl thiolate). The carboxylates are typically **silver** salts of long chain fatty acids and the n-alkyl thiolate is preferably 1-dodecanethiol. These **silver** (carboxylate-n-alkyl thiolate) particles can be used to formulate imaging forming compns. that are useful in aq. thermog. or photothermog. imaging elements.
- ST photog emulsion silver carboxylate alkyl thiolate particle photothermog
- IT Photographic emulsions

(heat-developable; **silver** (carboxylate-n-alkyl thiolate)

particles for photothermog. of thermog. imaging)

IT Surfactants

(nonionic; silver (carboxylate-n-alkyl thiolate)

particles for photothermog. of thermog. imaging)

IT Nanoparticles

(silver (carboxylate-n-alkyl thiolate) particles for

photothermog. of thermog. imaging)

IT 111-31-9, 1-Hexanethiol 112-55-0, 1-Dodecanethiol 112-85-6, Behenic acid **2885-00-9**, 1-**Octadecanethiol**

RL: TEM (Technical or engineered material use); USES (Uses)

(**silver** (carboxylate-n-alkyl thiolate) particles for photothermog. of thermog. imaging)

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

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- (2) Goffe; US 3666477 A 1972 CAPLUS
- (3) Lelental; US 6391537 B2 2002 CAPLUS
- (4) Voicu, R; Structure and Dynamics of Selectively Deuterated Self-Assembled Silver n-Octadecanethiolate Layered Materials P2266
- (5) Voicu, R; Thermal Behavior of a Self-Assembled Silver n-Dode-canethiolate Layered Material Monitored by DSC P2642
- L18 ANSWER 8 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN

Full Text

- AN 2003:798402 CAPLUS
- DN 139:311931
- ED Entered STN: 12 Oct 2003
- TI Metal coating of hair fibers for cosmetics
- IN Vic, Gabin; Livoreil, Aude; Giroud, Franck
- PA L'oreal, Fr.
- SO Fr. Demande, 18 pp.

CODEN: FRXXBL

- DT Patent
- LA French
- IC ICM A61K007-075
- CC 62-3 (Essential Oils and Cosmetics)

JP 2003300840 A

FAN.CNT 1

	9				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	FR 2838050	A1	20031010	FR 2002-4352	20020408
	FR 2838050	B1	20060714		
	CN 1449737	A	20031022	CN 2003-108449	20030331
	CN 1213719	С	20050810		
	BR 2003000873	A	20040817	BR 2003-873	20030403
	EP 1352630	A2	20031015	EP 2003-290860	20030407
	EP 1352630	A3	20040324		
	EP 1352630	В1	20060301		
	R: AT, BE, C	H, DE, DK	, ES, FR,	GB, GR, IT, LI, LU, NI	L, SE, MC, PT,
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	US 20030223944	A1	20031204	US 2003-407911	20030407

20031021 JP 2003-104420

20030408

JP 3759120 PRAI FR 2002-435 US 2002-372		B2 20060322 A 20020408 P 20020416
CLASS PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
FR 2838050	ICM IPCI IPCR	A61K007-075 A61K0007-075 [ICM, 7] A61K0008-00 [I,C*]; A61K0008-00 [I,A]; A61K0008-18 [I,C*]; A61K0008-18 [I,A]; A61K0008-19 [I,C*]; A61K0008-19 [I,A]; A61K0008-20 [I,A]; A61K0008-23 [I,A]; A61K0008-24 [I,A]; A61K0008-26 [I,A]; A61K0008-27 [I,A]; A61K0008-30 [I,C*]; A61K0008-31 [I,A]; A61K0008-34 [I,A]; A61K0008-35 [I,A]; A61K0008-37 [I,A]; A61K0008-46 [I,A]; A61K0008-64 [I,A]; A61K0008-72 [I,C*]; A61K0008-73 [I,A]; A61K0008-89 [I,A]; A61K0008-891 [I,A]; A61Q0001-02 [I,C*]; A61Q0001-02 [I,A]; A61Q0005-00 [I,C*]; A61Q0005-00 [I,A]; A61Q0005-10 [I,C*]; A61Q0005-10 [I,A]; A61Q0005-12 [I,C*]; A61Q0005-12 [I,A] A61Q005/12; A61K008/19; A61K008/27; A61K008/46;
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EP 1352630	IPCI IPCR	A61Q005/00; A61Q005/10 A61K0008-19 [I,C]; A61K0008-30 [I,C]; A61Q0005-00 [I,C]; A61Q0005-10 [I,C]; A61Q0005-10 [I,A]; A61K0008-19 [I,A]; A61K0008-46 [I,A]; A61Q0005-00 [I,A] A61K0008-00 [I,C*]; A61K0008-00 [I,A]; A61Q0005-10 [I,A]; A61K0008-18 [I,C*]; A61K0008-18 [I,A]; A61K0008-19 [I,C]; A61K0008-19 [I,A]; A61K0008-20 [I,A]; A61K0008-23 [I,A]; A61K0008-24 [I,A]; A61K0008-26 [I,A]; A61K0008-27 [I,A]; A61K0008-30

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[I,C]; A61K0008-31 [I,A]; A61K0008-34 [I,A];
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                        [I,A]; A61K0008-64 [I,A]; A61K0008-72 [I,C*];
                        A61K0008-73 [I,A]; A61K0008-89 [I,A]; A61K0008-891
                        [I,A]; A61Q0001-02 [I,C*]; A61Q0001-02 [I,A];
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                        A61Q005/00; A61Q005/10
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                IPCI
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                 IPCR
                        A61K0008-19 [I,C*]; A61K0008-19 [I,A]; A61K0008-30
                        [I,C^*]; A61K0008-46 [I,A]; A61Q0005-12 [I,C^*];
                        A61Q0005-12 [I,A]
                 NCL
                        424/070.100; 510/119.000
                 ECLA
                        A61K008/19; A61K008/46; A61Q005/12
 JP 2003300840
                 IPCI
                        A61K0008-00 [I,A]; A61Q0005-00 [I,A]; A61K0008-18
                        [I,A]; A61Q0001-02 [I,A]
                 IPCR
                        A61K0008-00 [I,C*]; A61K0008-00 [I,A]; A61K0008-18
                        [I,C*]; A61K0008-18 [I,A]; A61K0008-19 [I,C*];
                        A61K0008-19 [I,A]; A61K0008-20 [I,A]; A61K0008-23
                        [I,A]; A61K0008-24 [I,A]; A61K0008-26 [I,A];
                        A61K0008-27 [I,A]; A61K0008-30 [I,C*]; A61K0008-31
                        [I,A]; A61K0008-34 [I,A]; A61K0008-35 [I,A];
                        A61K0008-37 [I,A]; A61K0008-46 [I,A]; A61K0008-64
                        [I,A]; A61K0008-72 [I,C*]; A61K0008-73 [I,A];
                        A61K0008-89 [I,A]; A61K0008-891 [I,A]; A61Q0001-02
                        [I,C*]; A61Q0001-02 [I,A]; A61Q0005-00 [I,C*];
                        A61Q0005-00 [I,A]; A61Q0005-10 [I,C*]; A61Q0005-10
                        [I,A]; A61Q0005-12 [I,C*]; A61Q0005-12 [I,A]
                        A61Q005/12; A61K008/19; A61K008/27; A61K008/46;
                        A61Q005/00; A61Q005/10
     The invention relates to a treatment process which confers cosmetic
AB
     properties on hair fibers. The process consists of treating the fibers
     with a metal salt in the presence of a reducing agent, directly on the
     fiber to form the corresponding free metal. Thus, a lock of hair after
     being shampooed, was dried and an aq. soln. of AgNO3 was applied onto the
     hair. After the addn. of NaBH4, the natural pigmented hair was dark, with
     metallic brilliance reflected on it.
ST
    metal salt hair cosmetic
ΙT
    Alcohols, biological studies
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (C1-4; metal treatment of hair fibers for cosmetics)
ΤТ
     Alkanes, biological studies
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (C5-10; metal treatment of hair fibers for cosmetics)
ΙT
     Polyelectrolytes
       Surfactants
        (amphoteric; metal treatment of hair fibers for cosmetics)
ΙT
     Fats and Glyceridic oils, biological studies
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (animal; metal treatment of hair fibers for cosmetics)
ΙT
     Surfactants
        (anionic; metal treatment of hair fibers for cosmetics)
ΙT
     Polyelectrolytes
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Surfactants

(cationic; metal treatment of hair fibers for cosmetics) ΙT Cosmetics (emollients; metal treatment of hair fibers for cosmetics) ΙT Sulfates, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (hydrogen; metal treatment of hair fibers for cosmetics) ΤТ Antifoaming agents Antiperspirants Cosmetics Hair Hair preparations Perfumes Pigments, nonbiological Preservatives Reducing agents Shampoos Sunscreens Thickening agents (metal treatment of hair fibers for cosmetics) Alkaline earth salts ΙT Bromates Carbonates, biological studies Disulfides Halides Nitrates, biological studies Paraffin oils Phosphates, biological studies Polymers, biological studies Polysiloxanes, biological studies Proteins Rare earth salts Sulfates, biological studies Thioethers Thiosulfates Transition metal salts RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (metal treatment of hair fibers for cosmetics) ΤТ Bisulfites Enzymes, reactions Sulfites Thiols, reactions Thioredoxins RL: RCT (Reactant); RACT (Reactant or reagent) (metal treatment of hair fibers for cosmetics) ΤТ Cosmetics (moisturizers; metal treatment of hair fibers for cosmetics) ΙT Surfactants (nonionic; metal treatment of hair fibers for cosmetics) ΤТ Peroxysulfates RL: RCT (Reactant); RACT (Reactant or reagent) (peroxymonosulfates; metal treatment of hair fibers for cosmetics) Alcohols, biological studies ΙT RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process); PYP (Physical process); BIOL (Biological study); PROC (Process); USES (Uses)

(polyhydric; metal treatment of hair fibers for cosmetics)

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Sulfonic acids, biological studies
ΙT
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
        (salts; metal treatment of hair fibers for cosmetics)
ΙT
     Sulfinic acids
     Thiols, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (salts; metal treatment of hair fibers for cosmetics)
ΙT
     Salts, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (thiol; metal treatment of hair fibers for cosmetics)
ΙT
    Lactones
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
        (thiolactones; metal treatment of hair fibers for cosmetics)
     Fats and Glyceridic oils, biological studies
ΤТ
     RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
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     64-17-5, Ethanol, biological studies 67-63-0, Isopropanol, biological
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     123-86-4, Butyl acetate 141-78-6, EtOAc, biological studies
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    Molybdenum, salts 7440-02-0D, Nickel, salts 7440-05-3D, Palladium,
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     7440-31-5D, Tin, salts 7440-32-6D, Titanium, salts 7440-33-7D,
     Tungsten, salts 7440-36-0D, Antimony, salts 7440-50-8D, Copper, salts
     7440-57-5D, Gold, salts 7440-66-6D, Zinc, salts 7440-74-6D, Indium,
           7758-89-6, Cuprous chloride 7761-88-8, Silver nitrate,
     biological studies 7775-41-9, Silver fluoride 7783-89-3,
     Silver bromate 7783-90-6, Silver chloride, biological
     studies 7783-96-2, Silver iodide 7785-23-1, Silver
     bromide 7787-70-4, Cuprous bromide 10025-98-6, Dipotassium palladium
     tetrachloride
                   10294-26-5, Silver sulfate 10294-28-7, Gold
     tribromide 16903-35-8 16923-58-3, Disodium hexachloroplatinate
    19045-66-0D, Thiocarbamic acid, salts 73506-93-1, Diethoxyethane RL: COS (Cosmetic use); PEP (Physical, engineering or chemical process);
     PYP (Physical process); BIOL (Biological study); PROC (Process); USES
     (Uses)
        (metal treatment of hair fibers for cosmetics)
ΤТ
     50-81-7, Ascorbic acid, reactions 53-57-6, NaDPH 58-68-4, NaDH
     68-11-1, Thioglycolic acid, reactions 77-92-9D, Citric acid, salts 106-51-4, 2,5-Cyclohexadiene-1,4-dione, reactions 123-31-9,
     Hydroquinone, reactions 280-64-8, 9-BBN
                                                1758-73-2, Formamidinesulfinic
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     7803-51-2, Phosphine
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     17836-88-3
                25895-60-7, Sodium cyanoborohydride 37318-49-3, Protein
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     RL: RCT (Reactant); RACT (Reactant or reagent)
        (metal treatment of hair fibers for cosmetics)
              THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT 10
(1) Anon; PATENT ABSTRACTS OF JAPAN 1991, V015(243), PC-0842
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- (2) Anon; PATENT ABSTRACTS OF JAPAN 1992, V016(509), PC-0997
- (3) Dong Sung Pharmaceuticals Co L; EP 1243249 A 2002 CAPLUS

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(4) Grollier, J; US 4971596 A 1990 CAPLUS
(5) Kanebo Ltd; JP 03077806 A 1991 CAPLUS
(6) Katsumi, M; JP 04187625 A 1992 CAPLUS
(7) Lapidus, H; US 4195972 A 1980
(8) Richez, H; US 1055355 A 1913 CAPLUS
(9) Wella Ag; DE 2806603 A 1979 CAPLUS
(10) Wella Ag; DE 29621557 U 1997
L18 ANSWER 9 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
    2003:737150 CAPLUS
ΑN
DN
    139:250305
    Entered STN: 19 Sep 2003
ED
    Invisible patch for the controlled delivery of cosmetic, dermatological,
ТΤ
    and pharmaceutical active ingredients onto the skin
    Shefer, Adi; Shefer, Samuel
ΙN
PΑ
    USA
    U.S. Pat. Appl. Publ., 17 pp., Cont.-in-part of U.S. Ser. No. 91,935.
SO
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DT
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    English
LA
IC
    ICM A61K031-715
    ICS A61K009-70
INCL 424449000; 514061000
    63-6 (Pharmaceuticals)
    Section cross-reference(s): 62
FAN.CNT 2
                     KIND DATE
                                    APPLICATION NO.
    PATENT NO.
                                                         DATE
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                                        ______
                                                              ______
                      A1 20030918 US 2003-376736
    US 20030175333
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A2
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            BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU,
            MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN,
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                       A2 20051214
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    JP 2006519263 T
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                      A2 20020306
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PRAI US 2002-91935
    US 2003-376736
    WO 2004-US6106
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CLASS
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PATENT NO.
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US 20030175333 ICM
                     A61K031-715
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                      A61K009-70
               INCL
                     424449000; 514061000
                     A61K0031-715 [ICM, 7]; A61K0009-70 [ICS, 7]
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AΒ The present invention relates to a patch for controlled topical or transdermal delivery of effective levels of cosmetic, dermatol., and pharmaceutical active ingredients onto the skin, hair follicles, and sebaceous glands, with minimal discomfort and ease of use. The patch can be transparent or clear and comprises a rate-controlling matrix layer. The matrix layer comprises water-sensitive, bioadhesive, film forming polymers, a water sol. oligomer, and a surfactant. The cosmetic, dermatol., and pharmaceutical active ingredients are sol. or dispersed in the matrix. The patch becomes tacky when wetted and adheres onto the skin. The adhesive properties of the patch are sufficient to maintain the patch in place on the skin for the recommended treatment period while allowing the patch to be readily removed without causing skin irritation or leaving adhesive residue on the skin. For example, an antibiotic patch contained polyvinyl alc. 50, PVP 1, polysorbate 20 5, Maltrin 180 10, lactitol 5, glycerin 10, and chloramphenicol 0.55%.

- ST patch bioadhesive polymer oligosaccharide **surfactant**; antibiotic patch PVA PVP polysorbate chloramphenicol
- IT Glycosides
 - RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(alkyl polyglycosides; invisible patches contg. bioadhesive polymers and **surfactants**)

```
(amphoteric; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (anionic; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Surfactants
        (cationic; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Essential oils
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (clove; invisible patches contq. bioadhesive polymers and
        surfactants)
    Hair preparations
ΙT
        (conditioners; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Cosmetics
        (depilatories; invisible patches contq. bioadhesive polymers and
        surfactants)
ΙT
     Acne
     Burn
     Dandruff
     Pruritus
     Rhus diversiloba
     Rhus toxicodendron
        (drugs for; invisible patches contq. bioadhesive polymers and
        surfactants)
    Alcohols, biological studies
ΤТ
     Amides, biological studies
     Esters, biological studies
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (ethoxylated; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
    Hair preparations
        (growth stimulants; invisible patches contg. bioadhesive polymers and
        surfactants)
     Vein, disease
ΤТ
        (hemorrhoid, drugs for; invisible patches contg. bioadhesive polymers
        and surfactants)
ΙT
     Syrups (sweetening agents)
        (hydrolyzed starch; invisible patches contq. bioadhesive polymers and
        surfactants)
ΤT
    Allergy inhibitors
     Aloe barbadensis
     Analgesics
     Anti-infective agents
     Anti-inflammatory agents
     Antibacterial agents
     Antibiotics
     Antiemetics
     Antihistamines
     Antimicrobial agents
     Antioxidants
     Antiperspirants
     Antitussives
     Antiviral agents
     Chelating agents
     Chemotherapy
```

Cholinergic antagonists

Deodorants

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Disinfectants
     Fungicides
     Hemostatics
     Immunomodulators
     Insecticides
     Radical scavengers
     Sunscreens
     Suntanning agents
     Vasoconstrictors
     Vasodilators
     Wound healing promoters
        (invisible patches contg. bioadhesive polymers and surfactants
        )
ΤТ
    Amine oxides
     Amino acids, biological studies
     Carbohydrates, biological studies
     Caseins, biological studies
     Flavonoids
     Gelatins, biological studies
     Glycerides, biological studies
     Lanolin
     Lecithins
     Oligosaccharides, biological studies
     Paraffin oils
     Peptides, biological studies
     Polyamides, biological studies
     Polyesters, biological studies
     Polyoxyalkylenes, biological studies
     Polyoxyalkylenes, biological studies
     Polysaccharides, biological studies
     Proteins
     Retinoids
     Vitamins
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (invisible patches contg. bioadhesive polymers and surfactants
        )
ΤТ
     Anesthetics
        (local; invisible patches contq. bioadhesive polymers and
        surfactants)
ΤТ
     Cosmetics
        (moisturizers; invisible patches contq. bioadhesive polymers and
        surfactants)
TΤ
     Surfactants
        (nonionic; invisible patches contg. bioadhesive polymers and
        surfactants)
ΙT
     Amines, biological studies
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polyamines, nonpolymeric; invisible patches contg. bioadhesive
        polymers and surfactants)
     Alcohols, biological studies
ΤТ
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (polyhydric, propoxylated; invisible patches contq. bioadhesive
        polymers and surfactants)
ΙT
     Quaternary ammonium compounds, biological studies
     RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
```

(polymers; invisible patches contg. bioadhesive polymers and **surfactants**)

IT Skin, disease

(rash, drugs for; invisible patches contg. bioadhesive polymers and **surfactants**)

IT Cosmetics

(skin-lightening; invisible patches contg. bioadhesive polymers and surfactants)

IT Drug delivery systems

(tapes; invisible patches contg. bioadhesive polymers and surfactants)

IT Cosmetics

(wrinkle-preventing; invisible patches contg. bioadhesive polymers and **surfactants**)

IT Surfactants

(**zwitterionic**; invisible patches contg. bioadhesive polymers and **surfactants**)

IT 36574-66-0D, N-coco acyl derivs.

RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(cocoamidopropylbetaine; invisible patches contg. bioadhesive polymers and **surfactants**)

- IT 68-26-8, Retinol 96-26-4, Dihydroxyacetone 814-71-1, Calcium thioglycolate 34452-51-2, Potassium thioglycolate
 RL: COS (Cosmetic use); BIOL (Biological study); USES (Uses) (invisible patches contg. bioadhesive polymers and surfactants
- 50-70-4, Sorbitol, biological studies 50-70-4D, Sorbitol, oligomers ΙT contg. 50-78-2, Aspirin 50-81-7, Vitamin C, biological studies 50-99-7D, Glucose, esters 50-99-7D, D-Glucose, oligomers contg. 55-56-1, Chlorhexidine 56-81-5, Glycerin, biological studies Glutamic acid, N-acyl derivs. 57-48-7D, Fructose, oligomers contg. 57-50-1D, Sucrose, esters 57-50-1D, Sucrose, oligomers contg. 57-55-6, Propylene glycol, biological studies 58-86-6D, Xylose, oligomers contg. 59-23-4D, Galactose, oligomers contg. 59-87-0, Nitrofurazone 60-54-8, 69-65-8D, Mannitol, oligomers contg. 69-72-7, Salicylic Tetracycline acid, biological studies 69-79-4D, Maltose, oligomers contg. Xylitol, oligomers contg. 106-11-6, Diethylene glycol monostearate 107-36-8D, Isethionic acid, cocoyl derivs. 108-46-3, Resorcinol, biological studies 108-95-2, Phenol, biological studies 114-07-8, Erythromycin 115-83-3, Pentaerythritol tetrastearate 144-55-8, Sodium bicarbonate, biological studies 151-21-3, Sodium lauryl sulfate, biological studies 404-86-4, Capsaicin 497-19-8, Sodium carbonate, biological studies 585-86-4D, Lactitol, oligomers contq. 585-88-6D, Maltitol, oligomers contg. 770-35-4, Phenoxyisopropanol 1338-41-6, Sorbitan monostearate 1406-18-4, Vitamin E 2216-51-5 3380-34-5, 3458-28-4D, D-Mannose, oligomers contg. 6284-40-8 Triclosan 7439-97-6, Mercury, biological studies 7440-22-4, **Silver**, biological studies 7553-56-2, Iodine, biological studies 8011-96-9, Calamine 8050-81-5, Simethicone 9000-01-5, Gum arabic 9002-89-5, Polyvinyl alcohol 9002-98-6 9003-05-8, Polyacrylamide 9003-39-8, Polyvinylpyrrolidone 9004-64-2, Hydroxypropyl cellulose 9005-25-8, Starch, biological studies 9005-25-8D, Starch, hydrolyzates 9005-64-5, Polysorbate 20 9011-13-6, Styrene-maleic anhydride copolymer 9011-16-9, Methyl vinyl ether-maleic anhydride copolymer 11099-07-3, Glycerin stearate 11111-12-9, Cephalosporin 11140-06-0, Glycerin palmitate 12694-22-3, Diglyceryl monostearate 13718-94-0D, Palatinose, oligomers contg. 15687-27-1, Ibuprofen 18323-44-9, Clindamycin 25322-68-3, Polyethylene glycol 25322-69-4 25655-41-8, Povidone iodine 26658-19-5, Sorbitan tristearate 27195-16-0, Sucrose distearate

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    68424-04-4, Polydextrose 71185-87-0, Hexaglyceryl tristearate
    75537-01-8, Gantrez S-97 95461-64-6, Decaglyceryl pentastearate
    99734-29-9, Tetraglyceryl tristearate 99880-64-5, Glyceryl dibehenate
    106392-12-5, Polyoxyethylene polyoxypropylene block copolymer
    RL: COS (Cosmetic use); THU (Therapeutic use); BIOL (Biological study);
    USES (Uses)
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       )
    56-75-7, Chloramphenicol 94-09-7, Benzocaine
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    RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
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DN
OREF 117:23503a,23506a
ED Entered STN: 04 Oct 1992
  Aqueous emulsion for temporary protection of silver and copper surfaces
    against tarnishing
IN Grossmann, Hermann
PA Doduco GmbH und Co. Dr. Eugen Duerrwaechter, Germany
    Eur. Pat. Appl., 6 pp.
    CODEN: EPXXDW
DT Patent
LA German
   ICM C23F011-16
TC
CC
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B1 19960320
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DE 1991-4124955 A 19910727
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                ECLA C23F011/16B
ES 2086471
                IPCI C23F0011-16 [ICM, 6]; C23F0011-10 [ICM, 6, C*]
               IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
               ECLA C23F011/16B
AΒ
    The emulsion of pH 1-10 (preferably 2-4) comprises a hydrophobic inhibitor
    of a C \ge 12 thioalc. with \ge 1 SH group and its ester 0.05-50
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(preferably 2-20), emulsifier 0.05-50 (2-20), and an anionic or
     nonionic surfactant \leq 2 (0.05-1 g/L). The emulsifier comprises
     an alkoxylated and preferably ethoxylated branched C4-20 alc., an alkyl or
     alkylphenyl ether of polyethylene glycol. Ag, Cu, and their alloys are
     treated with the emulsion at >T (m.p. of inhibitor), rinsed with H2O at
     <T, and dried with hot air. An example emulsion of pH 3 and suitable for
     treatment of Ag and Ag alloys contains octadecanethiol 0.5-30,
     polyethylene glycol alkyl ether 0.5-30, and SDS \leq 1 g/L H20.
    tarnishing inhibitor silver copper; thiol SDS tarnishing inhibitor
     silver; SDS thiol tarnishing inhibitor copper; polyethylene glycol ether
     tarnishing inhibitor
    Thiols, uses
     RL: USES (Uses)
        (corrosion inhibitors, for copper and silver, with
        emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol)
     Tarnishing
        (of silver and copper alloys, aq. emulsion for prevention of)
    Corrosion inhibitors
        (thiols, with emulsifiers of alkyl or alkyl Ph ether of polyethylene
        glycol)
    Alcohols, compounds
     RL: PROC (Process)
        (C8-16, ethoxylated, corrosion inhibitor emulsion contg., thiol, for
        copper and silver and their alloys)
     copper alloy, base
       silver alloy, base
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (tarnishing of, thiol inhibitor for)
     25322-68-3D, Polyethylene glycol, alkyl and alkylphenyl ethers 151-21-3,
     uses
     RL: PROC (Process)
        (corrosion inhibitor emulsion contg., thiol, for copper and
        silver and their alloys)
IT 2885-00-9, Octadecanethiol
     RL: PROC (Process)
        (corrosion inhibitors, for copper and silver, with
        emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol)
    7440-22-4, Silver, reactions 7440-50-8, Copper, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (tarnishing of, thiol inhibitor for)
L18 ANSWER 11 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN
Full Text
    1991:89162 CAPLUS
    114:89162
OREF 114:15093a,15096a
    Entered STN: 09 Mar 1991
     Silver metal liquidlike films (MELLFs). The effect of surfactants
    Yogev, D.; Efrima, S.
     Dep. Chem., Ben-Gurion Univ. Negev, Beer-Sheva, 84105, Israel
     Langmuir (1991), 7(2), 267-71
    CODEN: LANGD5; ISSN: 0743-7463
    Journal
    English
     66-4 (Surface Chemistry and Colloids)
     Section cross-reference(s): 73, 74
     The effects of {\bf surfactants} on the prodn. and stabilization of {\bf Ag} metal
     liquidlike films (MELLFs) were studied. The main role of the surfactant
     is in stabilizing the Ag MELLFs and improving their properties
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(reflectivity, "fluidity"). A variety of different surfactants were found to be active, and of those investigated, anionic fluoroalkyl surfactants seem to be the most effective. In the case of anionic surfactants, the countercation has a significant effect on the Ag MELLF, esp. if it is a surface-active agent in itself. The effects of the surfactants on the interfacial tension and their effect on the measured reflectivities of the MELLFs are discussed in the context of the interfacial colloidal model of Ag MELLFs. silver metal liquidlike film formation; surfactant effect metal liquidlike film; interfacial tension metal liquidlike film Films (metal liq.-like, surfactant effects on formation of) Interfacial tension (of surfactant solns., silver metal liq.-like film formation in relation to) Sulfonic acids, compounds RL: PRP (Properties) (perfluoroalkane, ammonium and potassium salts, surfactant effect of, on **silver** metal liq.-like film formation) Surfactants (silver metal liq.-like film formation in presence of) Carboxylic acids, compounds RL: PRP (Properties) (perfluoro, ammonium salts, surfactant effect of, on silver metal liq.-like film formation) 7440-22-4, Silver, uses and miscellaneous RL: USES (Uses) (liq.-like metal film formation by, surfactant effects on) 577-11-7 **2885-00-9**, 1-Octadecanethiol 9002-93-1, Triton X 100 52584-45-9, Monflor 31 57534-41-5, Zonyl FSN 60529-61-5, Monflor 32 67479-85-0, Zonyl FSC 67479-86-1, Zonyl FSP RL: PRP (Properties) (silver metal liq.-like film formation in presence of) L18 ANSWER 12 OF 12 CAPLUS COPYRIGHT 2009 ACS on STN Full Text AN 1987:33631 CAPLUS DN 106:33631 OREF 106:5655a,5658a ED Entered STN: 07 Feb 1987 Maleimide copolymer and thermoplastic resin prepared by using this copolymer IN Kimura, Atsushi; Toyooka, Yutaka; Kishida, Kazuo PA Mitsubishi Rayon Co., Ltd., Japan SO PCT Int. Appl., 41 pp. CODEN: PIXXD2 Patent Japanese ICM C08F002-18 ICS C08F212-04; C08L033-14; C08L035-06; C08L051-04 35-4 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 38 FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ______ _____ WO 8604337 A1 19860731 WO 1986-JP17 19860117 W: AU, US RW: DE, FR, GB, IT, NL

JP 61163903 A 19860724 JP 1985-4907

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AU 8653567
                        А
                                19860813
                                           AU 1986-53567
                                                                    19860117
     EP 208790
                                19870121 EP 1986-900840
                                                                    19860117
                         A1
        R: DE, FR, GB, IT, NL
     CA 1262299 A1
                                19891010
                                           CA 1986-518902
                                                                    19860923
PRAI JP 1985-4907
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                                19850117
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WO 1986-IP17
                         Α
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     WO 1986-JP17
                         А
                                19860117
CLASS
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 WO 8604337
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                       C08F002-18
                 ICS
                        C08F212-04; C08L033-14; C08L035-06; C08L051-04
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                 IPCI
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                        C08F0002-12 [I,C*]; C08F0002-18 [I,A]; C08F0222-00
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                        [I,C*]; C08F0222-40 [I,A]; C08L0035-00 [I,C*];
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                        [I,A]
                        C08F222/40; C08L035/06+B5; C08L051/04+B2
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                 IPCI
                        C08F0212-04 [ICS, 4]; C08F0212-00 [ICS, 4, C*];
                        C08F0002-00 [ICA, 4]
 JP 61174248
                 IPCI
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                        C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
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                        [I,C*]; C08L0007-00 [I,A]; C08L0021-00 [I,C*];
                        C08L0021-00 [I,A]; C08L0023-00 [I,C*]; C08L0023-00
                        [I,A]; C08L0033-02 [I,A]; C08L0033-18 [I,A];
                        C08L0033-24 [I,A]; C08L0035-00 [I,C*]; C08L0035-06
                        [I,A]; C08L0051-00 [I,C*]; C08L0051-00 [I,A];
                        C08L0051-02 [I,A]; C08L0051-04 [I,A]; C08L0101-00
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                        C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
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                        C08L0035-06 [I,A]; C08L0051-00 [I,C*]; C08L0051-04
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 EP 208790
                 IPCI
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                        C08F0212-04 [ICS, 4]; C08F0212-00 [ICS, 4, C*];
                        C08L0033-14 [ICS, 4]; C08L0033-00 [ICS, 4, C*];
                        C08L0035-06 [ICS, 4]; C08L0035-00 [ICS, 4, C*];
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                        C08L0035-06 [I,A]; C08L0051-00 [I,C*]; C08L0051-04
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                        C08F222/40; C08L035/06+B5; C08L051/04+B2
 CA 1262299
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                        C08L0025-02 [ICS, 4]; C08L0025-00 [ICS, 4, C*];
                        C08L0051-04 [ICS, 4]; C08L0051-00 [ICS, 4, C*]
                        C08F0212-00 [I,C*]; C08F0212-04 [I,A]; C08L0025-00
                 IPCR
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[I,C*]; C08L0025-02 [I,A]; C08L0051-00 [I,C*]; C08L0051-04 [I,A]

AB A maleimide polymer with excellent heat stability during high-temp. molding and giving a product with excellent resistance to discoloration, heat, and impact when blended with a graft rubber, is prepd. by polymn. of a monomer selected from an arom. vinyl monomer, an unsatd. nitrile, and Me methacrylate 50-95, a maleimide 5-50, and other monomers 0-30% in the presence of a Ca phosphate-based dispersing agent and a nonionic surfactant [RO(CH2CH2O)n]mPO(OA)3-m (R = C8-30 alkyl, aralkyl; A = H, metal; m = 1-3, n = 5-50). The process minimizes the scale formation of formed polymers on a reactor wall during polymn. Thus, a mixt. of arylonitrile 20, styrene 170, and N-phenylmaleimide 10 parts in 100 parts water contg. AIBN 0.1, tert-Bu benzoate 0.1, tert-dodecyl mercaptan 0.3, Gafac GB 520 0.003, and Ca3PO4 0.5 part was suspension-polymd. at

80° for 3 h and at 120° for 2 h to give polymer beads

280-290° to give a sample exhibiting yellowing index (at

(particle diam. 180 μ , glass-transition temp. 125°). During the polymn., no scale formation was obsd. A blend of 55 parts maleimide copolymer and 45 parts graft polymer from polybutadiene 50, acrylonitrile 15, and styrene 35 parts contg. Mg stearate 0.3, tris(nonylphenyl) phosphite 0.1, and Antage W 400 0.2 phr was injection-molded at

280°) 31, notched Izod impact strength 16.4 kg-cm/cm2, Rockwell

hardness (R) 102, and Vicat softening point 108°, with no **silver** streak formation, compared with 44, 16, 101, and 104, with **silver** streak formation, when a maleimide copolymer prepd. in the presence of poly(vinyl alc.) as a dispersing agent was used.

- phenylmaleimide copolymer suspension polymn; acrylonitrile copolymer suspension polymn; styrene copolymer suspension polymn; calcium phosphate dispersant suspension polymn; polyethylene glycol lauryl ether phosphate; nonionic phosphate surfactant suspension polymn; scale prevention suspension polymn dispersant; ABS blend maleimide copolymer molding; heat stability maleimide copolymer molding
- IT Plastics, molded

RL: USES (Uses)

(ABS polymer-maleimide-contg. polymers, heat- and impact-resistant, heat-stable)

IT Heat-resistant materials

(maleimide-contg. polymers, heat stability improvement of)

IT Dispersing agents

(polyalkylene glycol phosphate-tricalcium phosphate, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention)

IT Scale (coating)

(prevention of, on reactor wall during suspension polymn. of maleimide-contq. monomer mixts., dispersing agents for)

IT Polymerization

(suspension, of maleimide-contg. monomer mixts., dispersing agents for, for scale formation prevention)

IT 9003-56-9

RL: USES (Uses)

(phenylmaleimide copolymer blends, heat-stable, resistant to discoloration, heat and impact)

IT 31621-07-5P, Acrylonitrile-N-phenylmaleimide-styrene copolymer

94858-30-7P, Acrylonitrile- α -methylstyrene-N-phenylmaleimide-styrene copolymer 101482-57-9P, Acrylonitrilemethyl methacrylate-N-phenylmaleimide-styrene copolymer

RL: PREP (Preparation)

(prepn. of, by suspension polymn., dispersing agents for, for improved heat stability and scale prevention during polymn.)

IT 51811-79-1, Gafac RE 610

RL: USES (Uses)

(suspensing agents, Gafac RE 610, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.)

IT 35604-29-6, Gafac GB 520

RL: USES (Uses)

(suspension agent, Gafac GB 520, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.)

IT 7758-87-4, Tricalcium phosphate

RL: USES (Uses)

(suspension agent, in suspension polymn. of maleimide-contg. monomer mixts., for scale formation prevention during polymn.)

RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Anon; JP 4983785 A
- (2) Anon; JP 5495689 A
- (3) Anon; JP 57125242 A CAPLUS
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- (5) Anon; JP 58129043 A CAPLUS
- (6) Anon; JP 58206657 A CAPLUS
- (7) Anon; JP 59184243 A CAPLUS

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